

## 20.1

## DESCRIBING ACIDS AND BASES

## SECTION BENCH

## Abstract

- List the properties of acids and bases
- Name an acid or base without giving the formula

### Key Terms

- **Protein**
- **Carbohydrate**

## Part II Completion

After these straightforward instructions for filling, your understanding of the categories used here is clear and is reinforced by the authors. Blackboard uses the category "entity as person", which includes you as well.

Aqueous solutions of 1 are electrolytes. Acids react with metal 2 to produce 3 gas. 4 are compounds that react with acids to form 5 and a salt.

Acids produce \_\_\_\_\_ ions when dissolved in water. There is a standard system for naming acids. For example,  $\text{HBr}$  is called \_\_\_\_\_ acid, while  $\text{HNO}_3$  is called \_\_\_\_\_ acid.

Glases produce 12 compounds dissolved in water. Some compounds that are 13 are named in the same way as ionic compounds.

100

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1. *Journal of Management Studies*, 1997, 34, 1, 1-14.

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### Part B: True-False

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- \_\_\_\_\_ 11. A compound that contains hydrogen atoms will be an acid when dissolved in water.
- \_\_\_\_\_ 12. An acid contains one or more hydroxide ions.
- \_\_\_\_\_ 13. Chemists have a special name for many acids.

## Section 20 1 Describing Acids And Bases Answers

**Xiaolong Qi**



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