

DESCRIBING CHEMICAL REACTIONS

Section Review - Objectives

- Explain how to write a word equation
- Describe how to write a skeleton equation
- List the steps for writing a complete chemical equation.

Vocabulary

- chemical equation
- skeleton equation
- catalyst
- coefficients
- balanced equation

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

A chemical reaction can be concisely represented by a chemical ____ 1 _____. The substances that undergo a chemical change are the ____ 2 _____. The new substances formed in a chemical reaction are the ____ 3 _____. In accordance with the law of conservation of ____ 4 _____, a chemical equation must be balanced. When balancing an equation, you place ____ 5 _____ in front of reactants and products so that the same number of atoms of each ____ 6 _____ are on each side of the equation. An equation must never be balanced by changing the ____ 7 _____ in the chemical formula of a substance.

Special symbols are used to show the physical state of a substance in a reaction. The symbol for a liquid is ____ 8 _____; for a solid, ____ 9 _____; and for a gas, ____ 10 _____. A substance dissolved in water is designated ____ 11 _____. If a ____ 12 _____ is used to increase the rate of a chemical reaction, its formula is written above the arrow.

- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 13. In an equation, a substance is shown to be in the gaseous state by placing an upward-pointing arrow after its formula.
- _____ 14. The symbol Δ placed over the arrow in an equation means that heat is supplied to the reaction.
- _____ 15. Atoms are destroyed in a chemical reaction.
- _____ 16. A skeleton equation is not a balanced equation.

Part D Questions and Problems

24. Write a balanced equation for each of these chemical reactions. I

a. Aluminum reacts with aqueous hydrochloric acid to form hydrogen gas and aqueous aluminum chloride.

b. Acetylene gas (C_2H_2) burns in a welding torch with oxygen to form carbon dioxide gas and water vapor.

Section 1describing Chemical Reaction Answer Key

**Donald J. Wink, Sharon Fetzer-
Gislason, Sheila McNicholas**



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Prentice Hall Science Explorer: Chemical Interactions Pearson/Prentice Hall,Michael J. Padilla,2004-10 Holt Chemistry Ralph Thomas Myers,2004 **Describing Chemical Engineering Systems** William Edwin Ranz,1970

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excess of material and redundancy *Essentials of Chemical Reaction Engineering* H. Scott Fogler, 2011 Accompanying DVD ROM contains many realistic interactive simulations Passing the State Science Proficiency Tests Craig A. Wilson, 2013-12-05 Passing the State Science Proficiency Tests presents essential content for elementary and middle school teachers who want to improve their science content background enhance their classroom instruction or pass the state science proficiency tests This book addresses different aspects of the physical life and earth sciences Each chapter was written by a science education expert and includes review questions with an accompanying answer key This book will enhance the effectiveness and competency of any pre service or in service elementary or middle school teacher Physical and Chemical Changes Edward P. Ortleb, Richard Cadice, 1993-09-01 General chemistry information including everything from matter to radioactivity For grades 5 to 9 *The Practice of Chemistry* Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB *Chemistry* Trace Jordan, Neville R. Kallenbach, 2017 Chemistry The Molecules of Life offers chemical insights within the context of health pharmaceuticals and the function of biological molecules The contextualized presentation of topics gives students a broad introduction to chemistry and helps them to see the relevance of chemistry to their personal lives The Go-To Guide for Engineering Curricula, Grades 6-8 Cary I. Sneider, 2014-11-25 How to engineer change in your middle school science classroom With the implementation of the Next Generation Science Standards your students won't just be scientists they'll be engineers But that doesn't mean you need to reinvent the wheel Respected science educator Cary Sneider has done the groundwork for you collecting a full range of time tested curriculum materials to seamlessly weave engineering and technology concepts into your math and science lessons In this volume you'll find descriptions of instructional materials specifically created for and tested in middle school science classrooms Features include A handy table that takes you straight to the chapters most relevant to your needs In depth commentaries and illustrative examples that demystify engineering curricula at the middle school level A vivid picture of what each curriculum looks like in the classroom the learning goals it accomplishes and how it helps address the NGSS More information on the integration of engineering and technology into 21st century science classrooms and why it will make a difference One of the most well respected science educators in the country Cary Sneider was an NGSS Writing Team Leader and is an associate research professor at Portland State University This publication uses hands on explorations that impact students by getting them to think like an engineer It's also great for exploring the

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Environmental Science Bernard J. Nebel, Richard T. Wright, 1993 Revolving around the principles of sustainability this new edition sets out to provide students with a balanced complete treatment of environmental issues their scientific basis history and future Material is revised to reflect changing environmental understanding and issues

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