

# CHEMICAL REACTIONS

# **Practice Problems**

In your notebook, solve the following problems. Use the 3-step problem-solving approach you learned in Chapter I.

#### SECTION 11.1 DESCRIBING CHEMICAL REACTIONS

- Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.
- Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.
- Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.
- 4. Write a balanced equation for the production of HCl gas from its elements.
- Write a sentence that completely describes the chemical reaction represented by this balanced equation.

 $2HCl(aq) + CaCO_3(s) \rightarrow CO_2(g) + CaCl_2(aq) + H_2O(l)$ 

Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.

 $2Ag(s) + S(s) \rightarrow Ag_{s}S(s)$ 

- Write a balanced equation representing the formation of aqueous sulfuric acid from water and sulfur trioxide gas.
- 8. Write a balanced equation from this word equation.

aqueous silver nitrate + copper metal -- silver metal + aqueous copper nitrate

9. Write a balanced equation for the following word equation.

phosphorus + oxygen → tetraphosphorous decoxide

## SECTION 11.2 TYPES OF CHEMICAL REACTIONS

- Write a balanced equation representing the reaction of magnesium with oxygen gas to produce magnesium oxide.
- Write the balanced equation for the reaction that occurs between aluminum and fluorine.
- Write the balanced equation for the production of oxygen gas and potassium chloride from the decomposition of potassium chlorate.
- Write the balanced equation for the reaction between hydrochloric acid and calcium metal. The products are hydrogen gas and calcium chloride.
- Write the balanced equation for the combustion of propane (C<sub>3</sub>H<sub>8</sub>) to produce carbon dioxide and water vapor.
- Write the balanced equation for the reaction between iron(III) chloride and sodium hydroxide. The products are iron(III) hydroxide and sodium chloride.

# **Section 11 1 Describing Chemical Reaction Answer Key**

**Pierre Gaspard, Irene Burghardt** 

## **Section 11 1 Describing Chemical Reaction Answer Key:**

Fundamentals of Groundwater Franklin W. Schwartz, Hubao Zhang, 2024-01-18 Fundamentals of Groundwater A thoroughly updated classic on the fundamentals of groundwater The second edition of Fundamentals of Groundwater delivers an expert discussion of the fundamentals of groundwater in the hydrologic cycle and applications to contemporary problems in hydrogeology The theme of the book is groundwater broadly defined and it covers the theory and practice of groundwater from basic principles of physical and chemical hydrogeology to their application in traditional and emerging areas of practice This new edition contains extensive revisions including new discussions of human impacts on aquifers and strategies and concepts for sustainable development of groundwater It also covers the theory of groundwater flow including concepts of hydraulic head and the Darcy equation and ground water surface water interactions as well as geochemistry and contamination Readers will also find A thorough introduction to the techniques of water resource investigations and regional groundwater flow Comprehensive explorations of groundwater chemistry and its applications in regional characterization and assessments of health impacts Practical discussions of groundwater contamination and water sustainability more generally Fulsome treatments of newly emerged contaminants like PFAS pathogens agricultural contaminants methane arsenic uranium and redox processes Perfect for undergraduate and graduate students taking courses in hydrogeology groundwater geoscience applied geoscience and groundwater and contaminant processes Fundamentals of Groundwater also benefits environmental consultants geochemists engineers and geologists Science Explorer Physical Science Michael J. Chemical Reactions and Their Control on the Femtosecond Time Scale Pierre Padilla, Ioannis Miaculis, Martha Cyr, Gaspard, Irene Burghardt, 2009-09-09 Continuing the tradition of the Advances in Chemical Physics series Volume 101 Chemical Reactions and Their Control on the Femtosecond Time Scale details the extraordinary findings reported at the XXth Solvay Conference on Chemistry held at the Universite Libre de Bruxelles Belgium from November 28 to December 2 1995 This new volume discusses the remarkable opportunities afforded by the femtosecond laser focusing on the host of phenomena this laser has made it possible to observe Examining molecules on the intrinsic time scale of their vibrations as well as their dissociative motions and electronic excitations represents only part of a broadened scientific window made possible by the femtosecond laser The assembled studies with follow up discussions reflect the many specialties and perspectives of the Conference s 65 participants as well as their optimism concerning the breadth of scientific discovery now open to them The studies shed light on the laser s enhanced technical reach in the area of coherent control of chemical reactions as well as of more general quantum systems The theoretical fundamentals of femto chemistry the unique behavior of the femtosecond laser and a view toward future technological applications were also discussed Femtochemistry chemical reaction dynamics and their control Coherent control with femtosecond laser pulses Femtosecond chemical dynamics in condensed phases Control of quantum many body dynamics Experimental observation of laser control Solvent dynamics and

RRKM theory of clusters High resolution spectroscopy and intramolecular dynamics Molecular Rydberg states and ZEKE spectroscopy Transition state spectroscopy and photodissociation Quantum and semiclassical theories of chemical reaction rates A fascinating and informative status report on the cutting edge chemical research made possible by the femtosecond laser Chemical Reactions and Their Control on the Femtosecond Time Scale is an indispensable volume for professionals and students alike The femtosecond laser and chemistry s extraordinary new frontier of molecular motions observed on the scale of a quadrillionth of a second Research chemists have only tapped the surface of the spectacular reach and precision of the femtosecond laser a technology that has allowed them to observe the dynamics of molecules on the intrinsic time scale of their vibrations dissociative motions and electronic excitations Volume 101 in the Advances in Chemical Physics series Chemical Reactions and Their Control on the Femtosecond Time Scale details their extraordinary findings presented at the XXth Solvay Conference on Chemistry in Brussels The studies reflect the work in part of the Conference s 65 participants including many prominent contributors Together they shed light on the laser's enhanced technical range in the area of coherent control of chemical reactions as well as of more general quantum systems. The theoretical fundamentals of femtochemistry the unique behavior of the femtosecond laser and a view toward future technological applications were also discussed An exceptionally up to date examination of the chemical analyses made possible by the femtosecond laser Chemical Reactions and Their Control on the Femtosecond Time Scale is an important reference for professionals and students interested in enhancing their research capabilities with this remarkable tool From 1993 to 1996 she worked with Dr P Gaspard at the Universite Libre de Bruxelles Belgium on the application of new semiclassical techniques to elementary chemical reaction processes Water Quality Engineering Mark M. Benjamin, Desmond F. Lawler, 2013-06-13 Explains the fundamental theory and mathematics of water and wastewater treatment processes By carefully explaining both the underlying theory and the underlying mathematics this text enables readers to fully grasp the fundamentals of physical and chemical treatment processes for water and wastewater Throughout the book the authors use detailed examples to illustrate real world challenges and their solutions including step by step mathematical calculations Each chapter ends with a set of problems that enable readers to put their knowledge into practice by developing and analyzing complex processes for the removal of soluble and particulate materials in order to ensure the safety of our water supplies Designed to give readers a deep understanding of how water treatment processes actually work Water Quality Engineering explores Application of mass balances in continuous flow systems enabling readers to understand and predict changes in water quality Processes for removing soluble contaminants from water including treatment of municipal and industrial wastes Processes for removing particulate materials from water Membrane processes to remove both soluble and particulate materials Following the discussion of mass balances in continuous flow systems in the first part of the book the authors explain and analyze water treatment processes in subsequent chapters by setting forth the relevant mass balance for the process reactor geometry and

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behavior dynamically arising from spaciotemporal interactions between parts at the micro level Describes a simulation platform for characterizing resilience of CPS Complexity Challenges in Cyber Physical Systems has been written for researchers practitioners lecturers and graduate students in computer engineering who want to learn all about M S support to addressing complexity in CPS and its applications in today s and tomorrow s world Theoretical Biochemistry Leif A. Eriksson, 2001-02-19 Theoretical chemistry has been an area of tremendous expansion and development over the past decade from an approach where we were able to treat only a few atoms quantum mechanically or make fairly crude molecular dynamics simulations into a discipline with an accuracy and predictive power that has rendered it an essential complementary tool to experiment in basically all areas of science This volume gives a flavour of the types of problems in biochemistry that theoretical calculations can solve at present and illustrates the tremendous predictive power these approaches possess A wide range of computational approaches from classical MD and Monte Carlo methods via semi empirical and DFT approaches on isolated model systems to Car Parinello QM MD and novel hybrid QM MM studies are covered The systems investigated also cover a broad range from membrane bound proteins to various types of enzymatic reactions as well as inhibitor studies cofactor properties solvent effects transcription and radiation damage to DNA

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