

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.100 M volume 25.00 mL acid base 0.100 M

b. 0.100 M volume 25.00 mL acid base 0.100 M

c. 0.100 M pH 4.75 compound that releases acetate
0.100 M acid base 0.100 M acid

Acetic Acid (pKa = 4.75) titrated with NaOH

0.100 M 0.100 M 0.100 M
0.100 M 0.100 M 0.100 M

Species	Acid (pH, Base pH, or Salt pH)	Acids with M_1 ($M_1 = 0.10$)	Conductivity (μ or mS)
CH_3COOH	<u>4.75</u>	<u>1</u>	<u>1</u>
CH_3COO^-	<u>9.25</u>	<u>1</u>	<u>1</u>
NaCl	<u>7.00</u>	<u>1</u>	<u>1</u>
NaOH	<u>13.00</u>	<u>1</u>	<u>1</u>

Note Taking Guide Episode 1101 Answers

Karin Nielsen-Saines



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