

1 FOCUS

Objectives

- 1.3.1 **Describe** how Lavoisier transformed chemistry.
- 1.3.2 **Identify** three steps in the scientific method.
- 1.3.3 **Explain** why collaboration and communication are important in science.

Guide for Reading

Build Vocabulary

Word Forms When discussing the two types of variables, explain that one meaning of *manipulate* is “to manage or control” and one meaning of *response* is “to answer or act in turn.”

Reading Strategy

Use Prior Knowledge Before students read the section, ask them what they know about the scientific method.

2 INSTRUCT

Connecting to Your World

Have students read the opening paragraph and look at the photo. Ask, **What made Fleming different from other scientists who had seen this mold?** (Fleming recognized the importance of this discovery. He assumed that the mold had released a chemical that prevented the growth of bacteria.)

Alchemy

Relate

Students who have read the Harry Potter books will be familiar with the name of one alchemist, Nicolas Flamel (1330–1414), who wrote a book about the philosopher’s stone. Alchemists believed that the philosopher’s stone could change base metals into gold.

Guide for Reading



Key Concepts

- How did alchemy lay the groundwork for chemistry?
- How did Lavoisier help to transform chemistry?
- What are the steps in the scientific method?
- What role do collaboration and communication play in science?

Vocabulary

scientific method
observation
hypothesis
experiment
manipulated variable
responding variable
theory
scientific law

Reading Strategy

Building Vocabulary After you read this section, explain the difference between a theory and a scientific law.



20 Chapter 1

Connecting to Your World

In 1928, Alexander Fleming, a Scottish scientist, noticed that a bacteria he was studying did not grow in the presence of a yellow-green mold. Other scientists had made the same observation, but Fleming was the first to recognize its importance. He assumed that the mold had released a chemical that prevented the growth of the bacteria. That chemical was penicillin, which can kill a wide range of harmful bacteria. In 1945, Fleming shared a Nobel Prize for Medicine with Howard Florey and Ernst Chain, who led the team that isolated penicillin. In this section you will study the methods scientists use to solve problems.



Alchemy

The word *chemistry* comes from alchemy. Long before there were chemists, alchemists were studying matter. Alchemy arose independently in many regions of the world. It was practiced in China and India as early as 400 B.C. In the eighth century, Arabs brought alchemy to Spain, from where it spread quickly to other parts of Europe.

Alchemy had a practical side and a mystical side. Practical alchemy focused on developing techniques for working with metals, glass, and dyes. Mystical alchemy focused on concepts like perfection. Because gold was seen as the perfect metal, alchemists were searching for a way to change other metals, such as lead, into gold. Although alchemists did not succeed in this quest, the work they did spurred the development of chemistry.

Alchemists developed the tools and techniques for working with chemicals. Alchemists developed processes for separating mixtures and purifying chemicals. They designed equipment that is still used today, including beakers, flasks, tongs, funnels, and the mortar and pestle in Figure 1.15. What they did not do was provide a logical set of explanations for the changes in matter that they observed. That task was left for chemists to accomplish.

Figure 1.15 A bowl-shaped mortar and a club-shaped pestle are used to grind or crush materials such as herbs, spices, and paint pigments. The mortar and pestle in the photograph is made of porcelain, which is a hard material.

Section Resources

Print

- [Guided Reading and Study Workbook, Section 1.3](#)
- [Core Teaching Resources, Section 1.3 Review](#)
- [Transparencies, T5–T6](#)
- [Laboratory Manual, Lab 1](#)
- [Small-Group Chemistry Laboratory Manual, Lab 1](#)

Technology

- [Interactive Textbook with ChemASAP, Assessment 1.3](#)
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Prentice Hall Chemistry Chapter 3 Study Guide

Xiaolong Qi



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Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry John McMurry, Susan McMurry, 1986 **Prentice Hall Chemistry**, 2000 *Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry* McMurry, Susan McMurry, 1990 [Study Guide and Solutions Manual for Organic Chemistry](#) Susan McMurry, 1992 John McMurry's best selling text presents organic chemistry in a new edition that is up to date beautifully written visually striking and pedagogically sound Described by many of its users as an eminently teachable text McMurry sets the standard in the field The writing style has received almost universal acclaim from its users McMurry introduces new concepts only as needed and immediately illustrates them with concrete examples And wherever possible he ties material together with brief reviews overviews and reaction summaries The result is a text that helps students mentally organize the material a text that helps them understand concepts not just memorize facts and a text that helps them make sense of the voluminous amount of material they encounter in the study of organic chemistry McMurry uses a simple but important polar reaction the addition of HBr to an alkene as the lead off reaction to illustrate the general principles of organic reactions Users of former editions found this an excellent choice because of its relative simplicity no prior knowledge of chirality or kinetics is required and its importance as a polar reaction on a common functional group that offers students the key to understanding hundreds of thousands of ionic reactions By selecting this particular model McMurry is able to offer an unusually early presentation of organic reactions *A Concise Guide to Improving Student Learning* Diane Cummings Persellin, Mary Blythe Daniels, 2023-07-03 This concise guidebook is intended for faculty who are interested in engaging their students and developing deep and lasting learning but do not have the time to immerse themselves in the scholarship of teaching and learning Acknowledging the growing body of peer reviewed literature on practices that can dramatically impact teaching this intentionally brief book Summarizes recent research on six of the most compelling principles in learning and teaching Describes their application to the college classroom Presents teaching strategies that are based on pragmatic practices Provides annotated bibliographies and important citations for faculty who want to explore these topics further This guidebook begins with an overview of how we learn covering such topics such as the distinction between expert and novice learners memory prior learning and metacognition The body of the book is divided into three main sections each of which includes teaching principles applications and related strategies most of which can be implemented without extensive preparation The applications sections present examples of practice across a diverse range of disciplines including the sciences humanities arts and pre professional programs This book provides a foundation for the reader explore these approaches and methods in his or her teaching **An Introduction to Archaeological Chemistry** T. Douglas Price, James H. Burton, 2010-10-17 Archaeological chemistry is a subject of great importance to the study and methodology of archaeology This comprehensive text covers the subject with a full range of case studies materials and research methods With twenty

years of experience teaching the subject the authors offer straightforward coverage of archaeological chemistry a subject that can be intimidating for many archaeologists who do not already have a background in the hard sciences With clear explanations and informative illustrations the authors have created a highly approachable text which will help readers overcome that intimidation Topics covered included Materials rock pottery bone charcoal soils metals and others Instruments microscopes NAA spectrometers mass spectrometers GC MS XRF XRD Case Studies Provenience Sediments Diet Reconstruction Past Human Movement Organic Residues The detailed coverage and clear language will make this useful as an introduction to the study of archaeological chemistry as well as a useful resource for years after that introduction

Polymer Science Study Guide Gerald S. Kirshenbaum, 1973 Computational Chemistry Errol G. Lewars, 2016-09-20
This is the third edition of the successful text reference book that covers computational chemistry It features changes to the presentation of key concepts and includes revised and new material with several expanded exercises at various levels such as harder questions for those ready to be tested in greater depth this aspect is absent from other textbooks in the field Although introductory and assuming no prior knowledge of computational chemistry it covers the essential aspects of the subject There are several introductory textbooks on computational chemistry this one is as in its previous editions a unique textbook in the field with copious exercises and questions and solutions with discussions Noteworthy is the fact that it is the only book at the introductory level that shows in detail yet clearly how matrices are used in one important aspect of computational chemistry It also serves as an essential guide for researchers and as a reference book **Chemistry for Degree Students B.Sc. (Honours) Semester II, 1/e (As per CBCS)** Madan R.L., 2022 This textbook has been designed to meet the needs of B Sc Honours Second Semester students of Chemistry as per the UGC Choice Based Credit System CBCS Maintaining the traditional approach to the subject this textbook lucidly explains the basics of Organic and Physical Chemistry Important topics such as alkanes alkenes alkynes stereochemistry aliphatic hydrocarbons thermochemistry chemical thermodynamics and chemical equilibrium are aptly discussed to give an overview of organic and physical chemistry Laboratory work has also been included to help students achieve solid conceptual understanding and learn experimental procedures **ENC Focus**, 2003 **NASA EP.** United States. National Aeronautics and Space Administration, 1969 Mathematics and Science for Students with Special Needs Eisenhower National Clearinghouse for Mathematics and Science Education, 2003 *Prentice Hall Science Explorer* Michael J. Padilla, 2002 Sedimentology of Aqueous Systems Cristiano Poletto, Susanne M. Charlesworth, 2010-02-05 Sediments in aqueous systems are of increasing interest to academics researchers practitioners and stakeholders around the world This book not only covers the characteristics of the sediments themselves but also their physico chemical impact on aquatic habitats and subsequent management implications There is a strong focus on methods and instrumentation for collecting data and monitoring of environmental sediment quality and as a result a wide range of environments are considered from urban areas to freshwater estuaries and marine ecosystems The chapters have been

written by international specialists in the field ensuring a good breadth of examples experiences and case studies throughout This book will appeal to a broad spectrum of interests from geographers to engineers and environmental scientists and at undergraduate to post graduate and academic researcher levels Physical Chemistry Robert J. Silbey, Robert A. Alberty, George A. Papadantonakis, Mounji G. Bawendi, 2022-06-15 Ever since Physical Chemistry was first published in 1913 it has remained a highly effective and relevant learning tool thanks to the efforts of physical chemists from all over the world Each new edition has benefited from their suggestions and expert advice The result of this remarkable tradition is now in your hands *Resources for Teaching Middle School Science* Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-03-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and

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Commonly Asked Questions in Thermodynamics Marc J. Assael, Geoffrey C. Maitland, Thomas Maskow, Urs von Stockar, William A. Wakeham, Stefan Will, 2022-08-05 CRC Press is pleased to introduce the new edition of *Commonly Asked Questions in Thermodynamics* an indispensable resource for those in modern science and engineering disciplines from molecular science engineering and biotechnology to astrophysics Fully updated throughout this edition features two new chapters focused on energy utilization and biological systems This edition begins by setting out the fundamentals of thermodynamics including its basic laws and overarching principles It provides explanations of those principles in an organized manner using questions that arise frequently from undergraduates in the classroom as the stimulus These early chapters explore the language of thermodynamics the first and second laws statistical mechanical theory measurement of thermodynamic quantities and their relationships phase behavior in single and multicomponent systems electrochemistry and chemical and biochemical reaction equilibria The later chapters explore applications of these fundamentals to a diverse set of subjects including power generation with and without fossil fuels for transport industrial and domestic use heating decarbonization technologies energy storage refrigeration environmental pollution and biotechnology Data sources for the properties needed to complete thermodynamic evaluations of many processes are included The text is designed for readers to dip into to find an answer to a specific question where thermodynamics can provide some if not all of the answers whether in the context of an undergraduate course or not Thus its readership extends beyond conventional technical undergraduates to practicing engineers and also to the interested lay person who seeks to understand the discourse that surrounds the choice of particular technological solutions to current and future energy and material production problems

Environmental Chemistry Stanley Manahan, Stanley E. Manahan, 2009-12-17 The field of environmental chemistry has evolved significantly since the publication of the first edition of *Environmental Chemistry* Throughout the book's long life it has chronicled emerging issues such as organochloride pesticides detergent phosphates stratospheric ozone depletion the banning of chlorofluorocarbons and greenhouse warming D

Study Guide Robert Rohli, Thomas Schmidlin, 2003-07 The *Study Guide* includes additional learning objectives a complete chapter outline critical thinking exercises problems and short essay work using actual figures from the text and a self test with an answer key in the back

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