

**Periodic Trends**

1. Explain the following trends:

- Atomic Radius
- Electronegativity
- Ionization Energy

2. Order the following elements by increasing atomic radius:

Li, Na, K, Rb, Cs, Fr

3. Order the following elements by increasing electronegativity:

F, Cl, Br, I, At

4. Order the following elements by increasing ionization energy:

Li, Na, K, Rb, Cs, Fr

5. Order the following elements by increasing metallic character:

Li, Na, K, Rb, Cs, Fr

6. Order the following elements by increasing non-metallic character:

F, Cl, Br, I, At

7. Order the following elements by increasing atomic mass:

Li, Na, K, Rb, Cs, Fr

8. Order the following elements by increasing atomic number:

Li, Na, K, Rb, Cs, Fr

9. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

10. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

**Periodic Trends Test Study Guide**

**Section 1: Atomic Structure and Properties of the Periodic Table**

1. What are the three main categories of elements in the periodic table?

- Metals
- Non-metals
- Metalloids

2. List the following elements in order of increasing atomic radius: Li, Na, K, Rb, Cs, Fr.

3. List the following elements in order of increasing electronegativity: F, Cl, Br, I, At.

4. List the following elements in order of increasing ionization energy: Li, Na, K, Rb, Cs, Fr.

5. List the following elements in order of increasing metallic character: Li, Na, K, Rb, Cs, Fr.

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9. List the following elements in order of increasing atomic weight: Li, Na, K, Rb, Cs, Fr.

10. List the following elements in order of increasing atomic weight: Li, Na, K, Rb, Cs, Fr.

**Section 2: Periodic Trends**

1. What are the three main categories of elements in the periodic table?

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10. List the following elements in order of increasing atomic weight: Li, Na, K, Rb, Cs, Fr.

**Section 3: Periodic Trends**

1. What are the three main categories of elements in the periodic table?

- Metals
- Non-metals
- Metalloids

2. List the following elements in order of increasing atomic radius: Li, Na, K, Rb, Cs, Fr.

3. List the following elements in order of increasing electronegativity: F, Cl, Br, I, At.

4. List the following elements in order of increasing ionization energy: Li, Na, K, Rb, Cs, Fr.

5. List the following elements in order of increasing metallic character: Li, Na, K, Rb, Cs, Fr.

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10. List the following elements in order of increasing atomic weight: Li, Na, K, Rb, Cs, Fr.

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**Toly K**



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## **Table of Contents Periodic Table And Chemistry Study Guide Answers**

1. Understanding the eBook Periodic Table And Chemistry Study Guide Answers
  - The Rise of Digital Reading Periodic Table And Chemistry Study Guide Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table And Chemistry Study Guide Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table And Chemistry Study Guide Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table And Chemistry Study Guide Answers
  - Personalized Recommendations
  - Periodic Table And Chemistry Study Guide Answers User Reviews and Ratings
  - Periodic Table And Chemistry Study Guide Answers and Bestseller Lists
5. Accessing Periodic Table And Chemistry Study Guide Answers Free and Paid eBooks
  - Periodic Table And Chemistry Study Guide Answers Public Domain eBooks
  - Periodic Table And Chemistry Study Guide Answers eBook Subscription Services
  - Periodic Table And Chemistry Study Guide Answers Budget-Friendly Options
6. Navigating Periodic Table And Chemistry Study Guide Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Table And Chemistry Study Guide Answers Compatibility with Devices
  - Periodic Table And Chemistry Study Guide Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Table And Chemistry Study Guide Answers
  - Highlighting and Note-Taking Periodic Table And Chemistry Study Guide Answers
  - Interactive Elements Periodic Table And Chemistry Study Guide Answers
8. Staying Engaged with Periodic Table And Chemistry Study Guide Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Periodic Table And Chemistry Study Guide Answers
- 9. Balancing eBooks and Physical Books Periodic Table And Chemistry Study Guide Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table And Chemistry Study Guide Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table And Chemistry Study Guide Answers
  - Setting Reading Goals Periodic Table And Chemistry Study Guide Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table And Chemistry Study Guide Answers
  - Fact-Checking eBook Content of Periodic Table And Chemistry Study Guide Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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