

### Chemical Formula

Talking about proportion molecules and relative of atoms present in a compound

Molecular  $\text{C}_{10}\text{H}_{12}\text{O}_4$  (molecular)

Formula represents the number of atoms of each  
in a single molecule

Empirical  $\text{C}_5\text{H}_6\text{O}_2$  (empirical)

Formula represents the simplest ratio of  
positive and negative ions in a formula  
unit

### Atomic Mass

The average mass of atoms present in a sample  
unit

### Mass

$m = \text{mass}$  unit mass

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# Note Taking Guide Episode 601 Chemical Formula Key

**WJ Hussar**



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