#### Nucleons:

- are particles occupying the NUCleus
- consist of t charged protons and wentral charge hentrons
- have almost 2000 times the Mass of 4cctrons
- are made up of quark and teptons

Nuclear Notation:

atomic wanber ()
= no. of protons (= Llarge)

 $^{27}_{13}AI$  has  $^{13}_{13}$  protons and  $^{19}_{13}$  neutrons, for a total of  $^{28}_{13}AI$  has  $^{28}_{13}$  protons and  $^{13}_{13}$  protons and  $^{13}_{13}$  neutrons, for a total of  $^{29}_{13}$  nucleons.

Forces Acting on Nucleons:

strong forces -

- · forces of (Attraction between nucleons
- are independent of the \_(ho(ges of the nucleon
- are short range (exist only between <u>CLOS=56</u> neighbors)

electrical force -

- · force of [204 15:0h between + charged protons
- are long range

When are nuclei unstable?

- a. large nuclei (Z>82) electrical forces of <u>Ft pulsion</u> are greater than strong forces of <u>A ttraction</u>
- b. wrong neutron : proton reabio

A radioactive isotope:

- . has an unstable nucleus
- · spontaneously emits a particle and Charge into another

CHEMISTRY: A Study of Matter

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