

# Relative Mass and the Mole

How can atoms be counted using a balance?

## Why?

Consider the following equation for a chemical reaction:  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

This can be interpreted as two molecules of hydrogen and one molecule of oxygen combining to form two water molecules. But how often do chemists limit their reactions to one or two molecules? Usually a reaction is done with an unimaginable number of molecules. How then do chemists know they have the right mix? The molecules need to be quickly counted! How do we count molecules? The answer is the unit called the **mole**. This activity will start by considering two egg farmers (a chicken farmer and a quail farmer). They produce such large numbers of eggs that they can't count them all individually, so they count in dozens of eggs in some cases, while in other cases they use mass. Weighing is often easier than counting!

## Model 1 – Eggs

| Chicken                      |                    | Quail                        |                    | Ratio of numbers of eggs | Ratio of masses of eggs |
|------------------------------|--------------------|------------------------------|--------------------|--------------------------|-------------------------|
| Number of eggs in the sample | Mass of the sample | Number of eggs in the sample | Mass of the sample |                          |                         |
| 1                            | 37.44 g            | 1                            | 2.34 g             | 1 : 1                    | 16 : 1                  |
| 10                           | 374.4g             | 10                           | 23.4g              | 1:1                      | 16:1                    |
| 438                          | 16394              | 438                          | 1.023g             | 1:1                      | 16:1                    |
| 1 dozen                      | <del>449.3g</del>  | 1 dozen                      | 28.08g             | 1:1                      | 16:1                    |
| 1 million                    | 37.                | 1 million                    | 2.340,000 g        | 1:1                      | 16:1                    |

- Consider the data in Model 1.
  - What is the mass of a standard chicken egg?  
37.44g
  - What is the mass of a standard quail egg?  
2.34g
  - Show mathematically how the 16:1 ratio of masses was calculated in the last column of Model 1.  

$$\frac{37.44}{2.34} = 16$$
- Use a calculator to complete the table in Model 1. Divide the work among group members. Reduce all ratios to the lowest whole numbers possible.

# Relative Mass And The Mole Dublin Schools

**E Durkheim**



## **Relative Mass And The Mole Dublin Schools:**

## **Relative Mass And The Mole Dublin Schools** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Relative Mass And The Mole Dublin Schools**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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