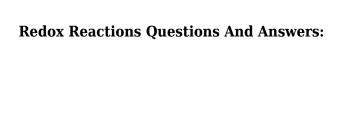
Redox Reactions

$$Z_{n(s)} + A_{9(nq)} \rightarrow Z_{n(nq)} + A_{9(s)}$$
 $M_{n0q} + F_{e(s)} \rightarrow F_{e^{3}} + M_{n^{2}} + A_{cidic}$
 $I + C_{103} \rightarrow I_{3} + C_{1}$
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Redox Reactions Questions And Answers

Lei Huang



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web the heat of reaction Δ hsoln is written after the products in units of kj mol rxn for example sodium hydroxide dissolves exothermically releasing 44 2 kilojoules per mole dissolved this process is represented as naoh s na aq oh aq Δ hsoln 44 2 kj mol rxn write equations to similarly represent the dissolving process for

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