* %yield could be in g or mol

$$\%$$
yield = $\frac{\text{actual yield}}{\text{theoretical yield}} \times 100$

$$= \frac{7.54 \text{g CaO}}{8.97 \text{g CaO}} \times 100 = 0.841 \times 100$$
$$= 8.97 \text{g CaO}$$
$$= 8.4.1 \%$$

Percent Yield Holt Chemfile Answers

D Keegan

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