



Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = 30.0 \text{ g/mol}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H₂O) gmm:

$$(2)(1.0) + (1)(16.0) = 18.0 \text{ g/mol}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = 88.9\%$$

$$\frac{45 \text{ g NO} \times \frac{1 \text{ mol NO}}{30 \text{ g NO}} \times \frac{4 \text{ mol NO}}{4 \text{ mol NO}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times 18 \text{ g H}_2\text{O}}{1} = 6.75 \text{ g H}_2\text{O}$$

Percent Yield Calculations Answers

Morris Hein, Susan Arena



Percent Yield Calculations Answers:

Unveiling the Power of Verbal Beauty: An Emotional Sojourn through **Percent Yield Calculations Answers**

In some sort of inundated with screens and the cacophony of quick transmission, the profound energy and mental resonance of verbal artistry usually disappear in to obscurity, eclipsed by the continuous onslaught of noise and distractions. However, set within the musical pages of **Percent Yield Calculations Answers**, a charming perform of fictional beauty that pulses with natural feelings, lies an wonderful journey waiting to be embarked upon. Penned by a virtuoso wordsmith, that mesmerizing opus books visitors on an emotional odyssey, softly revealing the latent potential and profound affect embedded within the intricate web of language. Within the heart-wrenching expanse with this evocative analysis, we will embark upon an introspective exploration of the book is key subjects, dissect their fascinating writing style, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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