

Algebra I: EOC Review Packet

Solving Linear Equations	<p>Solve the following equations for n.</p> <ol style="list-style-type: none">$2(2n + 3) + n = -5n + 1$$\frac{n}{4} - 3 = 7 - n$										
Solving Literal Equations	<ol style="list-style-type: none">Solve for E. $F + V = E + 2$Solve for r. $A = \pi r^2$Solve for x. $ax^2 + bx + c = 0$										
Exponent Properties	<ol style="list-style-type: none">Simplify: $9x^4y \cdot 3x^5y^{-8}$Fill in the blanks to make a true statement. <div style="border: 1px solid black; padding: 10px; width: fit-content; margin: 10px auto;">$\square x^{\square} y^{\square} \cdot \square x^{\square} y^{\square} = \frac{120y}{x^3}$</div>										
Linear Functions	<ol style="list-style-type: none">Find the rate of change between the points $(5, -4)$ and $(-10, -20)$Write an equation for the line represented by the points in the table below: <table border="1" style="margin: 10px auto;"><thead><tr><th>x</th><th>y</th></tr></thead><tbody><tr><td>-16</td><td>-22</td></tr><tr><td>-12</td><td>-19</td></tr><tr><td>-8</td><td>-16</td></tr><tr><td>-4</td><td>-13</td></tr></tbody></table>Choose all of the following options below that represent a line which is parallel to the line that contain the points: $\{(-10, 4); (-15, 2); (5, 10); (20, 16)\}$ <div style="margin-top: 10px;"><input type="checkbox"/> $y - 4 = \frac{2}{5}(x - 8)$ <input type="checkbox"/> $2x - 5y = 10$ <input type="checkbox"/> $5x + 2y = 7$ <input type="checkbox"/> $y = \frac{2}{5}x - 4$ <input type="checkbox"/> $-5x = 2y - 7$ <input type="checkbox"/> $x = \frac{5}{2}y + 15$</div>	x	y	-16	-22	-12	-19	-8	-16	-4	-13
x	y										
-16	-22										
-12	-19										
-8	-16										
-4	-13										
Systems of Equations	<ol style="list-style-type: none">Solve the systems:<ol style="list-style-type: none">$\begin{cases} 3x + 2y = -2 \\ y = -3x + 5 \end{cases}$$\begin{cases} 3x + 2y = -1 \\ 5x + 6y = -15 \end{cases}$										

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CK-12 Foundation

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web the relative electronegativities of the bonded atoms is $\text{H} < \text{C} < \text{Cl}$ and so the bond moments all point toward the Cl end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as BF_3 trigonal planar CH_4 tetrahedral PF_5 trigonal bipyramidal and SF_6

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