

Relative Mass and the Mole

How can atoms be counted using a balance?

Why?

Consider the following equation for a chemical reaction: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

This can be interpreted as two molecules of hydrogen and one molecule of oxygen combining to form two water molecules. But how often do chemists limit their reactions to one or two molecules? Usually a reaction is done with an unimaginable number of molecules. How then do chemists know they have the right mix? The molecules need to be quickly counted! How do we count molecules? The answer is the unit called the **mole**. This activity will start by considering two egg farmers (a chicken farmer and a quail farmer). They produce such large numbers of eggs that they can't count them all individually, so they count in dozens of eggs in some cases, while in other cases they use mass. Weighing is often easier than counting!

Model 1 – Eggs

Chicken		Quail		Ratio of numbers of eggs	Ratio of masses of eggs
Number of eggs in the sample	Mass of the sample	Number of eggs in the sample	Mass of the sample		
1	37.44 g	1	2.34 g	1 : 1	16 : 1
10	374.4g	10	23.4g	1:1	16:1
438	16394	438	1.023g	1:1	16:1
1 dozen	449.3g	1 dozen	28.08g	1:1	16:1
1 million	37.	1 million	2.340,000 g	1:1	16:1

- Consider the data in Model 1.
 - What is the mass of a standard chicken egg?
37.44g
 - What is the mass of a standard quail egg?
2.34g
 - Show mathematically how the 16:1 ratio of masses was calculated in the last column of Model 1.
$$\frac{37.44}{2.34} = 16$$
- Use a calculator to complete the table in Model 1. Divide the work among group members. Reduce all ratios to the lowest whole numbers possible.

Relative Mass And The Mole Key

Tom Jackson



Relative Mass And The Mole Key:

SuperSimple Chemistry DK,2020-06-09 A fantastic aid for coursework homework and studying for tests this comprehensive guide covers Next Generation Science Standards for grades 6 10 and will have you ready for tests and exams in no time Each topic is fully illustrated to support the information make the facts crystal clear and bring the science to life A large central image explains the idea visually and each topic is summed up on a single page helping children to quickly get up to speed and really understand how chemistry works Information boxes explain the theory with the help of simple graphics and for further studying a handy Key Facts box provides a simple summary you can check back on later With clear concise coverage of all the core topics SuperSimple Chemistry is the perfect accessible guide to chemistry for children supporting classwork and making studying for exams the easiest it s ever been Edexcel A Level Chemistry Student Book 1 Andrew Hunt,Graham Curtis,Graham Hill,2015-07-17 Exam Board Edexcel Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Endorsed by Edexcel Develop and assess your students knowledge and mathematical skills throughout A Level with worked examples practical assessment guidance and differentiated end of topic questions with this Edexcel Year 1 student book Identifies the level of your students understanding with diagnostic questions and a summary of prior knowledge at the start of the Year 1 Student Book Provides support for all 16 required practicals with various activities and questions along with a Practical chapter covering procedural understanding and key ideas related to measurement Mathematical skills are integrated throughout with plenty of worked examples including notes on methods to help explain the strategies for solving each type of problem Offers plenty of practice with Test Yourself Questions to help students assess their understanding and measure progress Encourages further reading and study with short passages of extension material Develops understanding with free online access to Test yourself Answers and an Extended Glossary Edexcel A level Chemistry Year 1 Student Book includes AS level Chemistry insights 'O' level Rex M. Heyworth,2007

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Table of Contents Relative Mass And The Mole Key

1. Understanding the eBook Relative Mass And The Mole Key
 - The Rise of Digital Reading Relative Mass And The Mole Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Relative Mass And The Mole Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Relative Mass And The Mole Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Relative Mass And The Mole Key
 - Personalized Recommendations

- Relative Mass And The Mole Key User Reviews and Ratings
- Relative Mass And The Mole Key and Bestseller Lists
- 5. Accessing Relative Mass And The Mole Key Free and Paid eBooks
 - Relative Mass And The Mole Key Public Domain eBooks
 - Relative Mass And The Mole Key eBook Subscription Services
 - Relative Mass And The Mole Key Budget-Friendly Options
- 6. Navigating Relative Mass And The Mole Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Relative Mass And The Mole Key Compatibility with Devices
 - Relative Mass And The Mole Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Relative Mass And The Mole Key
 - Highlighting and Note-Taking Relative Mass And The Mole Key
 - Interactive Elements Relative Mass And The Mole Key
- 8. Staying Engaged with Relative Mass And The Mole Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Relative Mass And The Mole Key
- 9. Balancing eBooks and Physical Books Relative Mass And The Mole Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Relative Mass And The Mole Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Relative Mass And The Mole Key
 - Setting Reading Goals Relative Mass And The Mole Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Relative Mass And The Mole Key
 - Fact-Checking eBook Content of Relative Mass And The Mole Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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