



Reinforcement

Chemical Changes

Read the passage and answer the following questions:



1. What is the chemical formula of hydrogen peroxide?

It is represented as the decomposition of the breaking of molecules of this chemical substance, the products are molecules of H_2O and the oxygen molecule O_2 which is released as a gas.

2. What is the chemical formula of hydrogen peroxide? Is it a molecular compound? Is it a ionic compound? Is it a metallic compound?

3. What is the physical state of hydrogen peroxide and the product?

Hydrogen peroxide is a colorless liquid with a strong odor. The products are water, which is a colorless liquid, and oxygen, which is a colorless gas. The reaction is exothermic, meaning it releases heat.

4. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much oxygen is produced?

10g grams



5. What is the chemical formula of the product? Is it a molecular compound?

6. What is the chemical formula of the reactant? Is it a molecular compound?

7. What is the physical state of the reactant and the product?

Reactant: Liquid
Product: Gas

8. What is the chemical formula of the product? Is it a molecular compound?

9. What is the chemical formula of the reactant? Is it a molecular compound?

Reactant: Liquid
Product: Gas

10. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much oxygen is produced?

Reinforcement Chemical Changes

David Kirk



Reinforcement Chemical Changes:

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