| Name | Period |
|----------|-----------|
| 17681880 | F ACREAGE |

Polyatomic Ion Worksheet

Place the charges on each ion - Write the Formula - Name the Compound

Remember - The cation always goes first.

| Exac | nple: NH | La*E | δt Γ' | Formula: NH ₄ I | Name: Ammonium Iodide |
|------|----------|------|--------|----------------------------|-----------------------|
| 1. | NH4 | 80 | 0 | Formula: | Name: |
| 2. | Ca | &c | CO_3 | Formula: | Name: |
| 3. | HCO_3 | St. | Na | Formula: | Name: |
| 4. | NO_3 | 80 | Mg | Formula: | Name: |
| 5. | к | Bi. | SO_4 | Formula: | Name: |
| 6. | Li | 86 | PO_4 | Formula: | Name: |

Finding the Formula from the Name

| Exam | pple: Lithium Chlor | ide lons: Li*1 & Cl*1 | Formula: LiCl |
|------|---------------------|-----------------------|---------------|
| 1. | Sodium Oxide | Ions: | Formula: |
| 2. | Potassium Bromide | lons: | Formula: |
| 3. | Calcium Nitride | lons: | Formula: |
| 4. | Beryllium Sulfide | lons: | Formula: |
| 5. | Magnesium Iodide | lons: | Formula: |
| 6 | Aluminum Oxide | Ions: | Formula: |
| 7. | Ammonium Bromide | lons: | Formula: |
| 8. | Calcium Bicarbonate | lons: | Formula: |
| 9. | Potassium Phosphate | Ions: | Formula: |
| LO. | Aluminum Sulfate | Ions: | Formula: |

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Tim McDowell

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