

# PERIODIC TABLE

## Atomic Properties of the Elements

**NIST**  
National Institute of  
Standards and Technology  
U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data					2										
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd					He										
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of <sup>133</sup> Cs																		13	14	15	16	17													
																		IIIA	IVA	VA	VIA	VIIA													
1		2		3		4		5		6		7		8		9		10	11	12	13	14	15	16	17	18									
H		He		Li		Be		B		C		N		O		F		Ne		Na		Mg		Al		Si		P		S		Cl		Ar	
Hydrogen 1.00794 (1.007825)		Helium 4.002602 (4.00260325315)		Lithium 6.941 (6.9408779)		Beryllium 9.012182 (9.0121831)		Boron 10.811 (10.8107)		Carbon 12.011 (12.010979)		Nitrogen 14.0064 (14.00643)		Oxygen 15.9994 (15.99903)		Fluorine 18.9984 (18.9984032)		Neon 20.1798 (20.179766)		Sodium 22.98976928 (22.98976928)		Magnesium 24.304 (24.304092)		Aluminum 26.9815386 (26.9815386)		Silicon 28.0855836 (28.0855836)		Phosphorus 30.973761998 (30.973761998)		Sulfur 32.06 (32.06)		Chlorine 35.45 (35.45)		Argon 39.948 (39.948)	
3		4		5		6		7		8		9		10		11		12		13		14		15		16		17		18		19		20	
Li		Be		B		C		N		O		F		Ne		Na		Mg		Al		Si		P		S		Cl		Ar		K		Ca	
Lithium 6.941 (6.9408779)		Beryllium 9.012182 (9.0121831)		Boron 10.811 (10.8107)		Carbon 12.011 (12.010979)		Nitrogen 14.0064 (14.00643)		Oxygen 15.9994 (15.99903)		Fluorine 18.9984 (18.9984032)		Neon 20.1798 (20.179766)		Sodium 22.98976928 (22.98976928)		Magnesium 24.304 (24.304092)		Aluminum 26.9815386 (26.9815386)		Silicon 28.0855836 (28.0855836)		Phosphorus 30.973761998 (30.973761998)		Sulfur 32.06 (32.06)		Chlorine 35.45 (35.45)		Argon 39.948 (39.948)		Potassium 39.0983 (39.0983)		Calcium 40.078 (40.078)	
11		12		13		14		15		16		17		18		19		20		21		22		23		24		25		26		27		28	
Na		Mg		Al		Si		P		S		Cl		Ar		K		Ca		Sc		Ti		V		Cr		Mn		Fe		Co		Ni	
Sodium 22.98976928 (22.98976928)		Magnesium 24.304 (24.304092)		Aluminum 26.9815386 (26.9815386)		Silicon 28.0855836 (28.0855836)		Phosphorus 30.973761998 (30.973761998)		Sulfur 32.06 (32.06)		Chlorine 35.45 (35.45)		Argon 39.948 (39.948)		Potassium 39.0983 (39.0983)		Calcium 40.078 (40.078)		Scandium 44.955912 (44.955912)		Titanium 47.88 (47.88)		Vanadium 50.9415 (50.9415)		Chromium 51.9961 (51.9961)		Manganese 54.938044 (54.938044)		Iron 55.845 (55.845)		Cobalt 58.933195 (58.933195)		Nickel 58.6934 (58.6934)	
37		38		39		40		41		42		43		44		45		46		47		48		49		50		51		52		53		54	
Rb		Sr		Y		Zr		Nb		Mo		Tc		Ru		Rh		Pd		Ag		Cd		In		Sn		Sb		Te		I		Xe	
Rubidium 85.4678 (85.4678)		Strontium 87.62 (87.62)		Yttrium 88.905848 (88.905848)		Zirconium 91.224 (91.224)		Niobium 92.90638 (92.90638)		Molybdenum 95.94 (95.94)		Technetium (98) (98)		Ruthenium 101.07 (101.07)		Rhodium 102.90550 (102.90550)		Palladium 106.42 (106.42)		Silver 107.8682 (107.8682)		Cadmium 112.411 (112.411)		Indium 114.818 (114.818)		Tin 118.710 (118.710)		Antimony 121.757 (121.757)		Tellurium 127.60 (127.60)		Iodine 126.90509 (126.90509)		Xenon 131.29 (131.29)	
55		56		57		58		59		60		61		62		63		64		65		66		67		68		69		70		71			
Cs		Ba		La		Ce		Pr		Nd		Pm		Sm		Eu		Gd		Tb		Dy		Ho		Er		Tm		Yb		Lu			
Cesium 132.90545196 (132.90545196)		Barium 137.327 (137.327)		Lanthanum 138.90547 (138.90547)		Cerium 140.116 (140.116)		Praseodymium 140.90766 (140.90766)		Neodymium 144.242 (144.242)		Promethium (145) (145)		Samarium 150.36 (150.36)		Europium 151.964 (151.964)		Gadolinium 157.25 (157.25)		Terbium 158.92535 (158.92535)		Dysprosium 162.500 (162.500)		Holmium 164.93033 (164.93033)		Erbium 167.259 (167.259)		Thulium 168.93401 (168.93401)		Ytterbium 173.054 (173.054)		Lutetium 174.967 (174.967)			
87		88		89		90		91		92		93		94		95		96		97		98		99		100		101		102		103			
Fr		Ra		Ac		Th		Pa		U		Np		Pu		Am		Cm		Bk		Cf		Es		Fm		Md		No		Lr			
Francium (223) (223)		Radium (226) (226)		Actinium (227) (227)		Thorium 232.0377 (232.0377)		Protactinium 231.036888 (231.036888)		Uranium 238.02891 (238.02891)		Neptunium (237) (237)		Plutonium (244) (244)		Americium (243) (243)		Curium (247) (247)		Berkelium (247) (247)		Californium (251) (251)		Einsteinium (252) (252)		Fermium (257) (257)		Mendelevium (258) (258)		Nobelium (259) (259)		Lawrencium (262) (262)			
104		105		106		107		108		109		110		111		112		113		114		115		116		117		118		119		120			
Rf		Db		Sg		Bh		Hs		Mt		Ds		Rg		Cn		Nh		Fl		Mc		Lv		Ts		Og		119		120			
Rutherfordium (261) (261)		Dubnium (262) (262)		Seaborgium (266) (266)		Bohrium (264) (264)		Hassium (277) (277)		Meitnerium (268) (268)		Darmstadtium (271) (271)		Roentgenium (272) (272)		Copernicium (285) (285)		Nihonium (284) (284)		Flerovium (289) (289)		Moscovium (288) (288)		Livermorium (292) (292)		Tennessine (294) (294)		Oganesson (294) (294)		119		120			

Atom Number

Ground-state Level

Symbol

Name

Atomic Weight

Ground-state Configuration

Ionization Energy (eV)

58

Ce

Cerium

140.116

[Xe]4f1 5s2

5.5037

Atomic Number: 58  
Ground-state Ionization Energy (eV): 5.5387

Symbol: **Ce**  
Name: Cerium  
Atomic Weight: 140.116

Ground-state Configuration: [Xe]4f<sup>1</sup>5d<sup>1</sup>6s<sup>2</sup>

\*Based upon <sup>12</sup>C. ( ) indicates the mass number of the longest-lived isotope.

For a description of the data, visit [physics.nist.gov/data](http://physics.nist.gov/data)

NIST SP 966 (September 2010)

# Periodix Table Study Packet

**Y Pai**



## Periodix Table Study Packet:

Know Your 'O' Level Chemistry - A Study Guide , Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement *MCAT Study Review Notes - MEGA PACK 900 Pages* E Staff, MCAT Study Review Notes MEGA PACK 900 Pages Prepare for the MCAT with this review notes mega pack 900 pages of best review notes from successful MCAT test takers Know all the important facts that you need to succeed on the MCAT From quick facts and mnemonics and everything in between is included in this mega pack Review all the important areas of science Be prepared to ace the test and get admitted into a medical school Content created by highly successful former MCAT test takers with in depth knowledge of what it takes to succeed in this exam ASAP Chemistry: A Quick-Review Study Guide for the AP Exam The Princeton Review, 2019-02-12 Looking for sample exams practice questions and test taking strategies Check out our extended in depth AP chem prep guide Cracking the AP Chemistry Exam LIKE CLASS NOTES ONLY BETTER The Princeton Review s ASAP Chemistry is designed to help you zero in on just the information you need to know to successfully grapple with the AP test No questions no drills just review Advanced Placement exams require students to have a firm grasp of content you can t bluff or even logic your way to a 5 Like a set of class notes borrowed from the smartest student in your grade this book gives you exactly that No tricks or crazy stratagems no sample essays or practice sets Just the facts presented with lots of helpful visuals Inside ASAP Chemistry you ll find Essential concepts terms and functions for AP Chem all explained clearly concisely Diagrams charts and graphs for quick visual reference A three pass icon system designed to help you prioritize learning what you MUST SHOULD and COULD know in the time you have available Ask Yourself questions to help identify areas where you might need extra attention A resource that s perfect for last minute exam prep and for daily class work Topics covered in ASAP Chemistry include Atomic structure Covalent bonding intermolecular forces Thermochemistry Acids bases and more **Study Guide to Accompany Basics for Chemistry** Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while

the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

**Safety Professional's Reference and Study Guide, Third Edition** W. David Yates, 2020-03-19 This new edition serves both as a reference guide for the experienced professional and as a preparation source for those desiring certifications It s an invaluable resource and a must have addition to every safety professional s library Safety Professional s Reference and Study Guide Third Edition is written to serve as a useful reference tool for the experienced practicing safety professional as well as a study guide for university students and those preparing for the Certified Safety Professional examination It addresses major topics of the safety and health profession and includes the latest version of the Board of Certified Safety Professional BCSP reference sheet a directory of resources and associations as well as state and federal agency contact information Additionally this new edition offers new chapters and resources that will delight every reader This book aids the prospective examination candidate and the practicing safety professional by showing them step by step how to solve each question formula listed on the BCSP examination and provide examples on how and when to utilize them

**Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice** Barron's Educational Series, Mark Kernion, Joseph A.

Mascetta, 2021-09-07 Barron s Science 360 Chemistry is your complete go to guide for everything chemistry This comprehensive guide is an essential resource for High school and college courses Homeschooling Virtual Learning Learning pods Inside you ll find Comprehensive Content Review Begin your study with the basic building block of chemistry and build as you go Topics include atomic structure chemical formulas electrochemistry the basics of organic chemistry and much more Effective Organization Topic organization and simple lesson formats break down the subject matter into manageable learning modules that help guide a successful study plan customized to your needs Clear Examples and Illustrations Easy to follow explanations hundreds of helpful illustrations and numerous step by step examples make this book ideal for self study and rapid learning Practice Exercises Each chapter ends with practice exercises designed to reinforce and extend key skills

and concepts These checkup exercises along with the answers and solutions will help you assess your understanding and monitor your progress Access to Online Practice Take your learning online for 50 practice questions designed to test your knowledge with automated scoring to show you how far you have come      *Safety Professional's Reference and Study Guide* W. David Yates,2017-12-12 While there are numerous technical resources available often you have to search through a plethora of them to find the information you use on a daily basis And maintaining a library suitable for a comprehensive practice can become quite costly The new edition of a bestseller Safety Professional's Reference and Study Guide Second Edition provides a single source reference that contains all the information required to handle the day to day tasks of a practicing industrial hygienist New Chapters in the Second Edition cover Behavior based safety programs Safety auditing procedures and techniques Environmental management Measuring health and safety performance OSHA's laboratory safety standard Process safety management standard BCSPs Code of Ethics The book provides a quick desk reference as well as a resource for preparations for the Associate Safety Professional ASP Certified Safety Professional CSP Occupational Health and Safety Technologist OHST and the Construction Health and Safety Technologist CHST examinations A collection of information drawn from textbooks journals and the author's more than 25 years of experience the reference provides as the title implies not just a study guide but a reference that has staying power on your library shelf      *Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone,Theodore O. Dolter,2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter      *Student Study Guide* Peter L. Kresan,Reed Mencke,Frank Press,Raymond Siever,2003-09-25 This reconceptualization of the text Understanding Earth reflects the fundamental changes in the field of physical geology over the past several years      *Digital SAT Study Guide Premium, 2024: 4 Practice Tests + Comprehensive Review + Online Practice* Brian W. Stewart,2023-08 An expert overview of the new Digital SAT how the exam is changing and what to expect from the digital interface on test day      *Digital SAT Study Guide Premium, 2025: 4 Practice Tests + Comprehensive Review + Online Practice* Barron's Educational Series,Brian W. Stewart,2024-07-02 Get ready for Digital SAT test day with Barron's

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**Study Guide for The Anatomy and Physiology Learning System** Edith Applegate, 2010-03-30 Designed to accompany The Anatomy and Physiology Learning System 4th Edition by Edith Applegate this study guide helps you learn and review basic A P concepts Each chapter emphasizes medical terminology with a set of key terms word parts clinical terms and abbreviations and then adds a variety of fun filled learning exercises review questions a quiz and a word puzzle The study guide corresponds to the textbook chapter for chapter Chapter learning objectives help you focus on the most important material Key concepts are defined on the first page of each chapter in the workbook Learning exercises for each chapter include short answer matching and diagrams to label and color Self quizzes allow you to measure your progress and understanding Fun and Games features end each chapter with a variety of engaging puzzles covering words and concepts A chapter summary provides a brief review of each chapter A chapter review provides questions for reinforcement and review of the concepts in each chapter

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**Level Study Guide - Chemistry Quite Easily Done** , Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011      **Understanding Earth Student Study Guide** Peter L. Kresan, Reed Mencke, 2006-05-03 The guide helps students prepare for lectures and exams with a heavy emphasis on utilizing the book's Web resources      **MCAT Complete; Medical College Admission Test Comprehensive Study Guide Physical Sciences; Biological Sciences; Verbal Reasoning** James Adams, Maryn Munley, Nathan Eisenbeis, Caroline Miller, Jesse Shetler, 2021-10-12 The MCAT The Medical College Admission Test MCAT is a standardized multiple choice examination designed to assess the examinee's problem solving critical thinking and knowledge of science concepts and principles prerequisite to the study of medicine Scores are reported in Physical Sciences Verbal Reasoning and Biological Sciences Study guide covers AAMC Association of American Medical Colleges content Physical Sciences Biological Sciences Verbal Reasoning Mathematics Concepts The Cell Chromosomes Reproduction Implantation Microorganisms Biochemistry Human Physiology The Heart The Lymphatic System GI Tract Musculoskeletal System Kidney Hormones Nerves Skin Genetics Populations and Evolution Elements Hund's Rule and Radiation The Periodic Table Covalent Bonds Molecular Shapes General Chemistry Kinetic Molecular Theory Phase Change Solutions Oxidation Numbers Entropy Acids and Bases Galvanic and Electrolytic Cells Carbon Stereochemistry Alkanes and Alkenes Hydrogen Bonding Alcohols Phenols Aldehydes and Ketones in Carboxylic Acids Ether Ammonia Amino Acids Carbohydrates Spectroscopy Separation and Distillation Vectors and Simple Motion Forces Circular and Projectile Motion Statics Center of Gravity Work and Energy Power and Momentum Stress and Strain Elasticity and Density Hydrostatic Pressure Fluids in Motion Electricity and Magnetism Plates Capacitors Voltage Batteries Resistors Magnetic Fields Waves and Periodic Motion Sound Waves Doppler Effect Simple Harmonic Motion Optics Mirrors 205 pages      **Study Guide to Physical Chemistry** Cybellium, Welcome to the forefront of knowledge with Cybellium your trusted partner in mastering the cutting edge fields of IT Artificial Intelligence Cyber Security Business Economics and Science Designed for professionals students and enthusiasts alike our comprehensive books empower you to stay ahead in a rapidly evolving digital world Expert Insights Our books provide deep actionable insights that bridge the gap between theory and practical application Up to Date Content Stay current with the latest advancements trends and best practices in IT AI Cybersecurity Business Economics and Science Each guide is regularly updated to reflect the newest developments and challenges Comprehensive Coverage Whether you're a beginner or an advanced learner Cybellium books cover a wide range of topics from foundational principles to specialized knowledge tailored to your level of expertise Become part of a global network of learners and professionals who trust Cybellium to guide their educational journey [www.cybellium.com](http://www.cybellium.com)      Chemistry in Everyday Life: A Study Guide Cybellium, 2024-10-26 Designed for professionals students and enthusiasts alike our comprehensive books empower you to stay ahead in a rapidly evolving digital world Expert Insights Our books provide deep actionable insights that bridge the gap between theory and practical application Up to Date Content Stay current with the

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