

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____

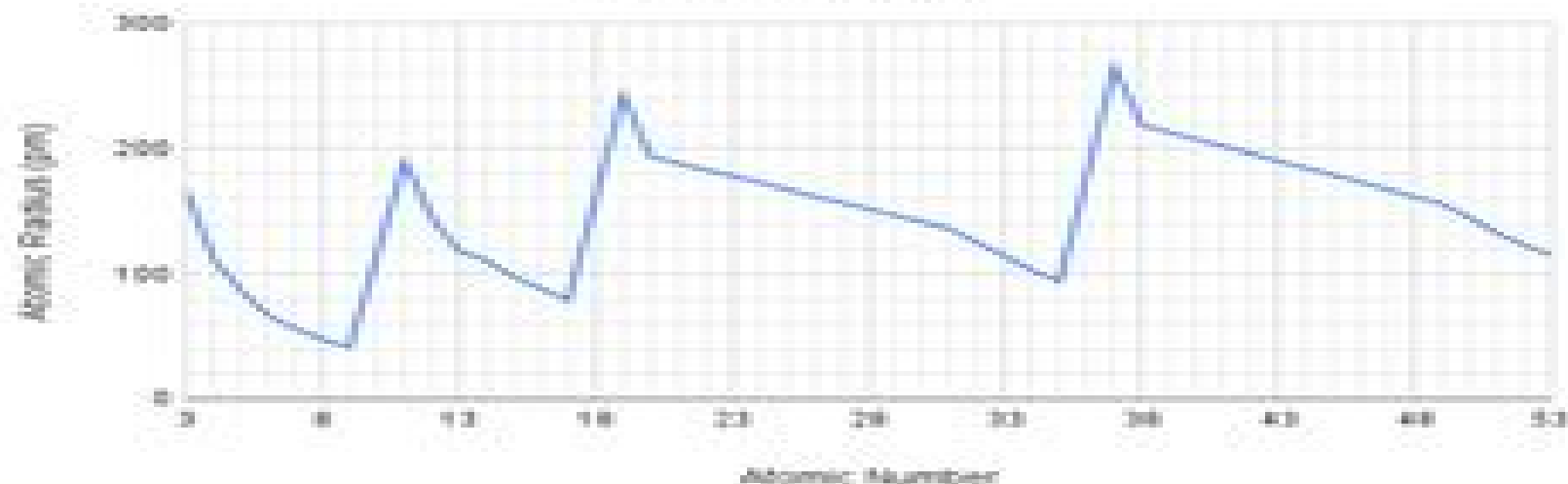
Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends, (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	175	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 52	125	Tin - 50	140
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

Atomic Radius



Periodic Trends Graphing Assignment

Ying-Ying Zheng

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Periodic Trends Graphing Assignment

1. Understanding the eBook Periodic Trends Graphing Assignment
 - The Rise of Digital Reading Periodic Trends Graphing Assignment
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Graphing Assignment
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Graphing Assignment
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Graphing Assignment
 - Personalized Recommendations
 - Periodic Trends Graphing Assignment User Reviews and Ratings
 - Periodic Trends Graphing Assignment and Bestseller Lists
5. Accessing Periodic Trends Graphing Assignment Free and Paid eBooks
 - Periodic Trends Graphing Assignment Public Domain eBooks
 - Periodic Trends Graphing Assignment eBook Subscription Services
 - Periodic Trends Graphing Assignment Budget-Friendly Options

6. Navigating Periodic Trends Graphing Assignment eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends Graphing Assignment Compatibility with Devices
 - Periodic Trends Graphing Assignment Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Graphing Assignment
 - Highlighting and Note-Taking Periodic Trends Graphing Assignment
 - Interactive Elements Periodic Trends Graphing Assignment
8. Staying Engaged with Periodic Trends Graphing Assignment
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Graphing Assignment
9. Balancing eBooks and Physical Books Periodic Trends Graphing Assignment
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Graphing Assignment
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Trends Graphing Assignment
 - Setting Reading Goals Periodic Trends Graphing Assignment
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Trends Graphing Assignment
 - Fact-Checking eBook Content of Periodic Trends Graphing Assignment
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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