

# THE PERIODIC TABLE OF ELEMENTS

## TEACHER ANSWER KEY *vocabulary practice*

### WORDS:

Period/row  
reactive

group/family  
nonreactive

metal  
conduct

nonmetal  
atomic mass

metalloid

### PART I: MAGIC SQUARE

Directions: Match the number of the statement best describing the word in the box and write it on the line below the word. IF they are all right, each column and row will total the same MAGIC NUMBER. Good luck!

reactive • <u>8</u>	metal • <u>3</u>	metalloid • <u>6</u>	<u>15</u>
atomic mass • <u>1</u>	group/family • <u>5</u>	nonmetal • <u>7</u>	<u>15</u>
conduct • <u>4</u>	nonreactive • <u>9</u>	period/row • <u>2</u>	<u>15</u>

TOTAL: 15      15      15

- Typically shiny, hard, solid elements that easily conduct heat and electricity. The majority of elements on the periodic table are this type of element.
- There are 7 of these on the PTE. Here you can see patterns such as increasing atomic number from left to right and decreasing reactivity of elements from left to right.
- On an element square, this number tells the total number of protons + neutrons.
- In science with relation to the periodic table, this is the ability to generate or create heat and/or electricity.
- Elements in this share similar properties, meaning they react and behave similarly to other elements found in the same column.
- An element that looks like a metal, but is a poor conductor of electricity.
- Elements that do not contain or resemble characteristics of metal. They are poor conductors of electricity.
- An element that easily and readily responds (joining, forming new substances) to other elements.
- An element that does not respond noticeably to the presence of other elements.

**OVER** →

# Periodic Table Vocabulary Answer Key

**M Woodhall**



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