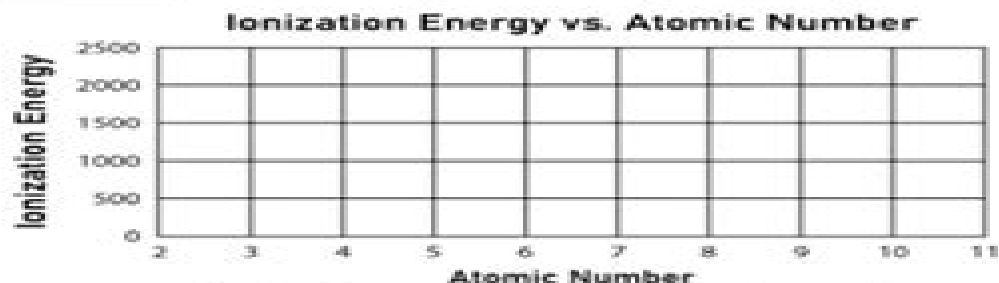


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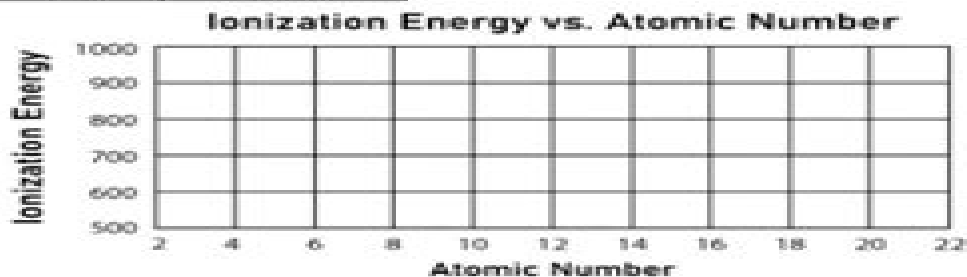
1. Using the ionization energy (IE) values of the elements in Period 2 listed below, make a line graph of IE vs. the atomic number.

Period 2					
Element	Atomic Number	IE (kJ/mol)	Element	Atomic Number	IE (kJ/mol)
Li	3	519	Li	3	519
Be	4	900	Be	4	900
B	5	799	B	5	799
C	6	1088	C	6	1088



- a) What trend do you notice for the ionization energies in Period 2?
- b) Explain why you observe this trend.
2. Make a scatter plot of ionization energy vs. atomic number for the first three atoms in Group 2A listed below.

Group 2A		
Element	Atomic Number	IE (kJ/mol)
Be	4	900
Mg	12	736
Ca	20	590



Reading Periodic Trend Graphing Questions

Scott C. Dulebohn



Reading Periodic Trend Graphing Questions:

Unveiling the Power of Verbal Artistry: An Emotional Sojourn through **Reading Periodic Trend Graphing Questions**

In a world inundated with screens and the cacophony of quick communication, the profound power and psychological resonance of verbal artistry usually disappear into obscurity, eclipsed by the regular assault of noise and distractions. However, located within the musical pages of **Reading Periodic Trend Graphing Questions**, a fascinating perform of literary brilliance that pulses with organic emotions, lies an remarkable trip waiting to be embarked upon. Written with a virtuoso wordsmith, this enchanting opus instructions visitors on a mental odyssey, softly revealing the latent possible and profound influence stuck within the intricate internet of language. Within the heart-wrenching expanse of the evocative analysis, we shall embark upon an introspective exploration of the book is key styles, dissect their captivating publishing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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Table of Contents Reading Periodic Trend Graphing Questions

1. Understanding the eBook Reading Periodic Trend Graphing Questions
 - The Rise of Digital Reading Reading Periodic Trend Graphing Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Reading Periodic Trend Graphing Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reading Periodic Trend Graphing Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Reading Periodic Trend Graphing Questions
 - Personalized Recommendations

- Reading Periodic Trend Graphing Questions User Reviews and Ratings
- Reading Periodic Trend Graphing Questions and Bestseller Lists
- 5. Accessing Reading Periodic Trend Graphing Questions Free and Paid eBooks
 - Reading Periodic Trend Graphing Questions Public Domain eBooks
 - Reading Periodic Trend Graphing Questions eBook Subscription Services
 - Reading Periodic Trend Graphing Questions Budget-Friendly Options
- 6. Navigating Reading Periodic Trend Graphing Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Reading Periodic Trend Graphing Questions Compatibility with Devices
 - Reading Periodic Trend Graphing Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reading Periodic Trend Graphing Questions
 - Highlighting and Note-Taking Reading Periodic Trend Graphing Questions
 - Interactive Elements Reading Periodic Trend Graphing Questions
- 8. Staying Engaged with Reading Periodic Trend Graphing Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reading Periodic Trend Graphing Questions
- 9. Balancing eBooks and Physical Books Reading Periodic Trend Graphing Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reading Periodic Trend Graphing Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Reading Periodic Trend Graphing Questions
 - Setting Reading Goals Reading Periodic Trend Graphing Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Reading Periodic Trend Graphing Questions
 - Fact-Checking eBook Content of Reading Periodic Trend Graphing Questions

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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