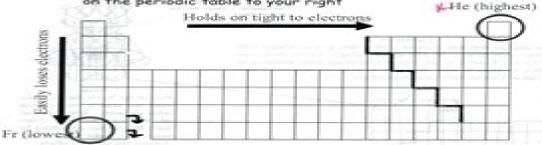
b. Why do elements get larger as you go down a group? Additional energy levels are added to the electron cloud as you go down a group. Land and no abmont

#### **Ionization Energy:**

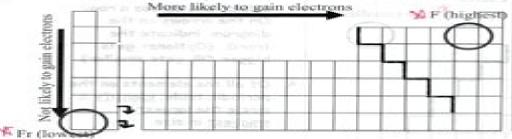
- 1. In simple terms, explain ionization energy. The energy required to remove an electron from a neutral atom.
- 2. Using the information in figure 5-15 (Pg 143) in your book. write the ionization energy for alkali metals & period #2 on the periodic table to your right



- On the arrows on the diagram. indicate the trend for ionization energy as you move down a column and across a row. (Options: easily loses an electron OR holds on tight to electrons)
- 4. Of all the elements on the periodic table, identify & circle the element with the highest & lowest ionization energy.
- Rank the following elements from highest to lowest ionization energy: a) Be, Ba, Mg = Be, Mg, Ba b) Li, F, B = F, B, Li
- 6. Why does it take less energy to remove an electron from a large atom compared to a small atom? Their electrons are closer to the nucleus, so the protons have a larger "pull" on them.

#### It takes less energy to remove an election from a large atom because Electronegativity: the effective nuclear charge decreases and electrons are

- 1. Explain Electronegativity: The ability of an atom to get electrons from other atoms.
- 2. Using the information in figure 5-20 (pg 151) in your book, write the electronegativity for the Halogens and period #2 on the periodic table below.



3. On the arrows on the diagram. fur Her indicate the trend for electronegativity as you move down a column and across a row. (Options: more likely to gain electrons, not likely to gain electrons)

WHILE BE

- 4. Of all the elements on the periodic table, identify & circle the element with the highest & lowest electronegativity.
- 5. Rank the following elements from most to least electronegative element: a) Be, Ba, Mg = Be, Mg, Ba b) Li, F, B - F, B, Li
- 6. Why are small atoms better at pulling electrons off of larger atoms than vice versa? Their electrons are closer to the nucleus, so the protons have a larger "pull" on the electrons in enough other atoms. The increased number of energy level (n) increases the distance over which nucleus must pull and therefore reduces the attraction for electrons

# **Pogil Cracking The Periodic Table Code Answer**

**J Dewey** 

**Pogil Cracking The Periodic Table Code Answer:** 

Pogil Cracking The Periodic Table Code Answer: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels enthralling the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the fascinating narratives that have charmed audiences this year. Pogil Cracking The Periodic Table Code Answer: Colleen Hoovers "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Uncover the Best: Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Pogil Cracking The Periodic Table Code Answer: Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and thrilling novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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