Name: Date: Period:

# POGIL-The Mole Concept

(How can atoms be counted?)

#### Why?

Consider the following equation for a chemical reaction:  $2H_2 + O_2 \rightarrow 2H_2O$ 

This can be interpreted as two molecules of hydrogen and one molecule of oxygen combining to form two water molecules. But how often do chemists limit their reactions to one or two molecules? Usually a reaction is done with an unimaginable number of molecules. How then do chemists know they have the right mix? The molecules need to be quickly counted! How do we count molecules? The answer is the unit called the **mole**. Weighing is often easier than counting!

# Model 1: The Balanced Equation

How can the coefficients in a chemical equation be interpreted?

Consider the reaction:

#### Key Questions

What are the coefficients for each of the following substances in the reaction?

#### Read This!

Coefficients tell you the relative amounts of products and reactants. A balanced chemical equation can be interpreted in terms of different quantities, including:

Numbers of particles (atoms, molecules, or formula units)

#### PARTICLES

Atoms Molecules Formula Units

- (if you have a single element) (if you have a covalent compound) (if you have an ionic compound)
- Mass
- Volume
- Moles

# **Pogil Introduction To Mole Concept**

**N Noddings** 

**Pogil Introduction To Mole Concept:** 

# Pogil Introduction To Mole Concept Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Pogil Introduction To Mole Concept**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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