Name _______ Pd

Unit 7: Balancing Chemical Reactions · Worksheet 2

Balance the following equations by inserting the proper coefficients.

- 1. $_2$ _SO₂ + $_1$ _O₂ \rightarrow $_2$ _SO₃
- 2. $2 \text{ CH4} + 3 \text{ O2} \rightarrow 2 \text{ CO} + 4 \text{ H2O}$
- 2 P + 3 Cl2 → 2 PCl3
- 2_CO + 1_O₂ → 2_CO₂
- 5. $_2$ CH4 + $_1$ O2 \rightarrow $_2$ CH3OH
- 6. $_2$ _Li + $_1$ _Br₂ \rightarrow $_2$ LiBr
- 7. $_2AI_2O_3 \rightarrow _4AI + _3O_2$
- 8. $_{2}$ Na + $_{2}$ H₂O \rightarrow $_{2}$ NaOH + $_{1}$ H₂
- 9. $_{6}$ _CO₂ + $_{6}$ _H₂O \rightarrow $_{1}$ _C₆H₁₂O₆ + $_{6}$ _O₂
- 10. _1_H2SO4 + _2_NaCl → _2_HCl + _1_Na2SO4

Two reactions used to get rid of sulfur dioxide, a pollutant from burning coal:

- 11. 3 H₂ + 1 SO₂ → 1 H₂S + 2 H₂O
- 12. $_2$ CaCO3 + $_2$ SO2 + $_1$ O2 \rightarrow $_2$ CaSO4 + $_2$ CO2
- 13. 2_AgNO₃ + 1_CaCl₂ → 2_AgCl + 1_Ca(NO₃)₂
- 14. 2 HCl + 1 Ba(OH)2 → 1 BaCl2 + 2 H2O
- 15. _1_H3PO4 + _3_NaOH → _1_Na3PO4 + _3_H2O
- 16. _1_Pb(NO3)2 + _2_KI → _1_PbI2 + _2_KNO3
- 17. $3 \text{ O}_1\text{ O} + 2 \text{ NH}_3 \rightarrow 1 \text{ N}_2 + 3 \text{ O}_1 + 3 \text{ H}_2\text{ O}_1$
- 18. $_1$ C₂H₅OH + $_3$ C₂ \rightarrow $_2$ CO₂ + $_3$ H₂O
- 19. $_2$ C2H6 + $_3$ O2 \rightarrow $_2$ CH3COOH + $_2$ H2O
- 20. $3NO_2 + 1H_2O \rightarrow 2HNO_3 + 1NO$

Rearranging Atoms Worksheet

Guillaume Favre



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Table of Contents Rearranging Atoms Worksheet

- 1. Understanding the eBook Rearranging Atoms Worksheet
 - The Rise of Digital Reading Rearranging Atoms Worksheet
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Rearranging Atoms Worksheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Rearranging Atoms Worksheet
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Rearranging Atoms Worksheet
 - Personalized Recommendations
 - Rearranging Atoms Worksheet User Reviews and Ratings
 - Rearranging Atoms Worksheet and Bestseller Lists
- 5. Accessing Rearranging Atoms Worksheet Free and Paid eBooks
 - Rearranging Atoms Worksheet Public Domain eBooks
 - Rearranging Atoms Worksheet eBook Subscription Services
 - Rearranging Atoms Worksheet Budget-Friendly Options

- 6. Navigating Rearranging Atoms Worksheet eBook Formats
 - o ePub, PDF, MOBI, and More
 - Rearranging Atoms Worksheet Compatibility with Devices
 - Rearranging Atoms Worksheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Rearranging Atoms Worksheet
 - Highlighting and Note-Taking Rearranging Atoms Worksheet
 - Interactive Elements Rearranging Atoms Worksheet
- 8. Staying Engaged with Rearranging Atoms Worksheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Rearranging Atoms Worksheet
- 9. Balancing eBooks and Physical Books Rearranging Atoms Worksheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Rearranging Atoms Worksheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Rearranging Atoms Worksheet
 - Setting Reading Goals Rearranging Atoms Worksheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Rearranging Atoms Worksheet
 - Fact-Checking eBook Content of Rearranging Atoms Worksheet
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

• Interactive and Gamified eBooks

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