Name	answer key	Date
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# TYPES OF CHEMICAL REACTIONS

Complete each of the following equations as needed to make it the type of reaction indicated.

Be sure to write each formula correctly.

Double replacement:

 $Na_2CrO_4 + PbCl_2$ 

2 Nacl + PbCrO<sub>4</sub>

2) Single replacement:

 $Cl_2 + 2 Nabr$ 

2 NaCl + Br2

3) Decomposition:

 $Mg(ClO_3)_2$ 

2 H<sub>2</sub>O

4) Synthesis:

 $2 H_2 + O_2$ 

2 Fe(OH)3 + 3 CaCl2

5) Double replacement:

3 Ca (OH)2 + 2 FeCl3

2 H<sub>2</sub>O

6) Single replacement:

 $Fe + Cu(NO_3)_2$ 

 $Fe(NO_3)_2 + Cu$ 

7) Decomposition:

 $2 \text{ Hg}_2 O$ 

 $4Hg + O_2$ 

8) Synthesis:

 $S + O_2$ 

 $50_2$ 

9) Double replacement:

AgNO<sub>3</sub> + KI

Agl + KNO<sub>3</sub>

10) Single replacement:

Cu + 2 AgNO<sub>3</sub>

 $2 \text{ Ag} + \text{Cu}(\text{NO}_3)_2$ 

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K. S. Verma

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