

Practice: pages 20–25

- 1 Answer is B. From experiments 1 and 3 the order with respect to Cl_2 is 1 (the rate quadruples as the concentration doubles. Therefore, in an experiment using 0.80 mol dm^{-3} of CO and 0.20 mol dm^{-3} of Cl_2 , the rate will be halved from the $6.4 \times 10^{-24} \text{ mol dm}^{-3} \text{ min}^{-1}$ obtained in experiment 1. (There is no need to work out the order with respect to CO.) [1]
- 2 Answer is B. The concentration halves every 120 s. So, the sequence of half-lives is $2.4 \rightarrow 1.2 \rightarrow 0.6 \rightarrow 0.3 \rightarrow 0.15$; that is four half-lives which is $4 \times 120 \text{ s}$. [1]
- 3 Answer is C. For a zero order reaction, the rate of decomposition is a constant. [1]
- 4 Answer is A. the decomposition is $\text{Rate} = k[\text{X}]^2$ [1]
From the data given, $0.016 = k[1]^2$ [1]
Therefore, $k = 0.016 \text{ s}^{-1}$. If $[\text{X}] = 0.25 \text{ mol dm}^{-3}$, $\text{rate} = 0.016[0.25]^2$
 $= 0.001 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$ [1]
- 5 Answer is B. The overall order is 2, so the units are $\frac{1}{\text{concentration} \times \text{s}}$. [1]
- 6 Answer is D. I_2 has an order of zero and is not involved in the rate-determining step. Therefore, it can be ignored. In D, $[0.1][2]$ has a value that is greater than $[0.1][1]$. [1]
- 7 Answer is C. The slow step + the fast step must give the overall equation. I must, therefore, be removed in the fast step in a reaction with G, and some J and K must be formed. [1]
- 8 Answer is B. The rate equation is $\text{Rate} = k[\text{halide}]$ so statements 1 and 2 are correct. Raising the temperature will increase the rate of reaction by allowing more molecules to exceed the activation energy, but the activation energy is fixed and will not change. [1]
- 9 Answer is D. Increasing the amount of ozone present does not change the order with respect to ozone. A large excess is not necessary to ensure enough ozone is present. The purpose of a large excess is to make sure that during the reaction its concentration is virtually constant. Therefore, only statement 3 is correct. [1]
- 10 Answer is A. Experiments 2 and 3 give the order with respect to X as zero. Experiments 1 and 2 give the order with respect to Y as 2. So statements 1 and 2 are correct. The rate constant is calculated from a set of results, for example, experiment 1 gives $0.08 = k[0.10]^2$.
So $k = 8.0 \text{ dm}^3 \text{ mmol}^{-1} \text{ s}^{-1}$ and statement 3 is, therefore, correct. [1]
- 11 a) $2\text{N}_2\text{O}_5(\text{g}) \rightarrow 4\text{NO}_2(\text{g}) + \text{O}_2(\text{g})$ [1]
b) i) For every 2 mol of N_2O_5 that reacts 4 mol of NO_2 is formed. Therefore, the rate of formation of NO_2 is twice as fast as the loss of N_2O_5 , i.e. $7.0 \times 10^{-4} \text{ mol dm}^{-3} \text{ s}^{-1}$. [1]
ii) For oxygen, the rate of formation will be $\frac{1}{2}$ the rate of loss of N_2O_5 ,
i.e. $1.75 \times 10^{-4} \text{ mol dm}^{-3} \text{ s}^{-1}$ [1]

Reaction Rate And Energy Reinforcement Answers Key

Michael Seilmaier



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and many others are also included No other reference source provides such comprehensive treatment of behavior modification history biography theory and application Thematic Coverage The first of the thematic volumes covers Adult Clinical Applications Adults are the most common population encountered by researchers clinicians and students and therefore more than 150 entries were needed to cover all necessary methods The second volume covers Child Clinical Applications in 140 entries One especially useful aspect of this volume will be the complications sections addressing what can go wrong in working with children This is an area often overlooked in journal articles on the subject Volume III Educational Applications addresses a range of strategies and principles of applied behavior analysis positive behavior support and behavior modification and therapy These entries focus on classroom and school contexts in which the instructional and behavioral interactions between teachers and their learners are emphasized Unique Easy to Follow Format Each of the volumes entries address a full range of mental health conditions and their respective treatments with the aim of providing systematic and scientific evaluation of clinical interventions in a fashion which will lend itself to the particular style of treatment common to behavior modification Major entries for specific strategies follow a similar format 1 Description of the Strategy 2 Research Basis 3 Relevant Target Populations and Exceptions 4 Complications 5 Case Illustration 6 Suggested Readings 7 Key Words Biographical sketches include the following 1 Birthplace and Date 2 Early Influences 3 Education History 4 Professional Models 5 Major Contributions to the Field 6 Current Work and Views 7 Future Plans Readership This encyclopedia was designed to enhance the resources available to students scholars practitioners and other interested social science readers The use of in text citations jargon and descriptions of research designs and statistics has been minimized making this an accessible comprehensive resource for students and scholars alike Academic and research librarians in the social sciences health and medicine will all find this an invaluable addition to their collections Key Features Three thematic volumes and over 430 total entries Five anchor articles in each volume provide context on major issues within the field Key words and lists of suggested readings follow each entry Contributions by internationally renowned authors from England Germany Canada Australia New Zealand and the United States Volume Editors Volume I Adult Clinical Applications Michel Hersen Johan Rosqvist Pacific University Volume II Child Clinical Applications Alan M Gross Ronald S Drabman University of Mississippi Volume III Educational Applications George Sugai Robert Horner University of Oregon Advisory Board Thomas M Achenbach Ph D Department of Psychiatry University of Vermont Stewart W Agras M D Department of Psychiatry Behavioral Science Stanford University School of Medicine David H Barlow Ph D ABPP Center of Anxiety and Related Disorders Boston University Alan S Bellack Ph D ABPP Department of Psychiatry University of Maryland School of Medicine Edward B Blanchard Ph D Department of Psychology University of Albany SUNY James E Carr Ph D Department of Psychology Western Michigan University Anthony J Cuvo Ph D Rehabilitation Institute Southern Illinois University Gerald C Davison Ph D Department of Psychology University of Southern California Eric F Dubow Ph D Psychology Department Bowling Green State

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Publications of the National Bureau of Standards 1975 Catalog United States. National Bureau of Standards,1976

Oncology Rehabilitation E-Book Deborah Doherty,Chris Wilson,Lori Boright,2022-07-23 Coverage of physical therapy patient management includes acute care outpatient and multidisciplinary clinical settings along with in depth therapeutic management interventions Content on the continuum of cancer care addresses the primordial primary secondary tertiary and quaternary stages in prevention and

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