

Cracking the Periodic Table Code

Why aren't the elements listed in alphabetical order?

Why?

As charts go, the periodic table is a bit odd. It's not square. Large portions of the table appear to be missing at the top. It's not organized alphabetically so elements can be found easily. But to a chemist, the periodic table is a very powerful tool. The periodic table is organized by properties, both chemical and physical. Those properties relate to the electronic structure of the atoms of each element. In today's activity, you are going to study how the ground state electron configurations and structure of atoms are related to the shape and organization of the periodic table.

Blank Periodic Table

A blank periodic table grid with 18 columns and 7 rows. The label $3p^3$ is located in the third row, eighth column. The grid is designed for a chemistry assignment, with the label $3p^3$ indicating the electron configuration for the element in that position.

1. Circle the number of the team your group has been assigned.
- Team 1: H, Li, Na, K, Rb, Cs
Team 2: Fe, Co, Ni, Ru, Rh, Pd
Team 3: Ne, Ar, Kr, Xe, Rn
Team 4: F, Cl, Br, I, At
Team 5: Be, Mg, Ca, Sr, Ba
Team 6: N, P, As, Sb, Bi
Team 7: Ti, V, Cr, Zr, Nb, Mo
Team 8: honors only Er, Tm, Yb, Fm, Md, No
Team 9: honors only Sm, Tb, Ho, Pu, Bk, Es

Pogil Cracking The Periodic Table Code

George Graybill



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