

- ☐ As more protons are added, the more tightly the surrounding electrons are drawn inwards.
- ☐ As more neutrons are added, the more tightly the surrounding electrons are drawn inwards.
- ☐ As more electrons are added, the more tightly they draw themselves towards the nucleus.
- ☐ As the mass of the nucleus rises, the more tightly the surrounding electrons are drawn inwards.

- ☐ It decreases.
- ☐ It decreases, then increases.
- ☐ It remains constant.
- ☐ It increases, then decreases.
- ☐ It increases.

A periodic table of elements. Each element is represented by a box containing its symbol, atomic number, and name. The table is color-coded by groups: alkali metals (red), alkaline earth metals (orange), transition metals (yellow), post-transition metals (light green), metalloids (green), nonmetals (blue), and noble gases (purple).

- ☐ Argon (Ar)
- ☐ Phosphorus (P)
- ☐ Sodium (Na)
- ☐ Aluminum (Al)
- ☐ Magnesium (Mg)

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