

Name: _____ Date: _____

Periodic Table Scavenger Hunt

Search your periodic table for the answers to these questions.

- | | |
|--------------------|--|
| <u>Silicon</u> | 1. Which element is number 14 on the periodic table? |
| <u>Cf</u> | 2. What is the element symbol for californium? |
| <u>83</u> | 3. How many protons are in an atom of bismuth? |
| <u>Noble Gas</u> | 4. To which element group does argon belong? |
| <u>Cadmium</u> | 5. Which element would you expect to have a higher mass: cadmium or zinc? |
| <u>12.01</u> | 6. What is the atomic mass of carbon? |
| <u>Lanthanides</u> | 7. What do you call the element series from atomic number 57-71? |
| <u>Gold</u> | 8. Which element has a symbol that starts with a letter different from the first one in its name: aluminum, copper, gold, rhenium? |
| <u>Hydrogen</u> | 9. Which element has the lowest atomic mass? |
| <u>Ruthenium</u> | 10. What is the first element with an atomic mass greater than 100? |
| <u>Aluminum</u> | 11. What is the first basic metal on the periodic table? |
| <u>False</u> | 12. True or false: Tin and antimony are in the same element group. |
| <u>Francium</u> | 13. What is the heaviest alkali metal? |
| <u>12</u> | 14. How many protons are in an atom of magnesium? |
| <u>Silicon</u> | 15. Which of the following is not a nonmetal: sulfur, oxygen, silicon, nitrogen? |
| <u>Tungsten</u> | 16. What is the name of the element with the symbol W? |
| <u>Palladium</u> | 17. Which element has an atomic mass of 106.42? |
| <u>Halogen</u> | 18. Astatine belongs to which element group: nonmetal, halogen, noble gas? |
| <u>Barium</u> | 19. What is the element with the symbol Ba? |
| <u>I or O</u> | 20. Name a letter never used in any element symbol? |

Periodic Table Scavenger Hunt Key For

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Inquiry: The Key to Exemplary Science Robert Yager,2009-06-17 **Content Area Reading and Learning** Diane Lapp,James Flood,Nancy Farnan,2016-11-18 How can teachers make content area learning more accessible to their students This text addresses instructional issues and provides a wealth of classroom strategies to help all middle and secondary teachers effectively enable their students to develop both content concepts and strategies for continued learning The goal is to help teachers model through excellent instruction the importance of lifelong content area learning This working textbook provides students maximum interaction with the information strategies and examples presented in each chapter This book is organized around five themes Content Area Reading An Overview The Teacher and the Text The Students The Instructional Program School Culture and Environment in Middle and High School Classrooms Pedagogical features in each chapter include a graphic organizer a chapter overview Think Before Think While and Think After Reading Activities which are designed to integrate students previous knowledge and experience with their new learnings about issues related to content area reading literacy and learning and to serve as catalysts for thinking and discussions This textbook is intended as a primary text for courses on middle and high school content area literacy and learning **Who's the New Kid in Chemistry?** John D. Butler,2013-12-12 Who s the New Kid in Chemistry offers an unprecedented look at student engagement and teacher best practices through the eyes of an educational researcher enrolled as a public high school student Over the course of seventy nine consecutive days John D Butler participates in and observes Rhode Island 2013 Teacher of the Year Jessica M Waters s high school chemistry class documenting his experiences as they unfold Who s the New Kid in Chemistry is a compelling example of what can be accomplished when an educational researcher and teacher collaborate in the classroom This work includes a discussion on flexible homework assignments data driven instruction and thirty teacher best practices This book is an invaluable resource for teachers across all content areas masters and doctoral research method classes and future Teachers of the Year A Guide for Using the Red Pony in the Classroom Mari Lu Robbins,1994 Answer Key for The Mystery of the Periodic Table Staff of Christ the King Books,2018-02 The Basics of the Periodic Table Leon Gray,2013-12-15 A sweeping history of both the discovery and classification of elements and the development of the modern periodic table Included are discussions of the discovery of matter atoms atomic structure molecules compounds ions and isotopes as well as the first identifications of the 118 and counting elements and the various ways they have been classified and organized by prominent scientists up to the present day periodic table Instruction in how to read the periodic table is accompanied by examinations of the various groups of elements their location on the table and their properties and practical uses This text strongly supports Common Core Standards for the reading of scientific and technical texts and accounts and furnishes ample opportunities to summarize cite evidence and analyze connections between ideas individuals and events **Atoms, Molecules & Elements: Patterns In the Periodic Table Gr. 5-8** George Graybill,2015-10-01 This

is the chapter slice Patterns In the Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

The Periodic Table in Minutes Dan Green,2016-09-06 An icon of science the Periodic Table defines the fundamental chemistry of everything in the universe In this compact yet comprehensive guide Dan Green outlines the history development and workings of the table shows how its design reflects and illuminates the organisation of all matter and even explains what it has to tell us about the chemistry of distant stars and of our own bodies Contents include an individual entry for every known element detailing properties uses and key data and sections on the patterns and groups of the famous table as well as explanations of basic chemistry concepts such as elements and compounds atomic structure chemical bonds reactions and radioactivity amongst many others *Mastering the Periodic Table* Linda Trombley,Faye Williams,2000 Whether students are studying chemistry biology or other sciences the periodic table is a vitally important tool These 50 word games puzzles and other creative activities unlock the nature of the various elements while explicating periodicity atomic structure element groups and more Complete teacher support includes background information answer keys and materials lists

Atoms, Molecules & Elements: The Periodic Table Gr. 5-8 George Graybill,2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

The Periodic Table Adrian Dingle,Dan Green,2014 The original Basher Science made even better *Mystery of the Periodic Table* Benjamin D Wiker,2003-04-18 Leads the reader on a delightful and absorbing journey through the ages on the trail of the elements of the Periodic Table as we know them today He introduces the young reader to

people like Von Helmholtz, Boyle, Stahl, Priestly, Cavendish, Lavoisier and many others all incredibly diverse in personality and approach who have laid the groundwork for a search that is still unfolding to this day. The first part of Wiker's witty and solidly instructive presentation is most suitable to middle school age while the later chapters are designed for ages 12-13 and up with a final chapter somewhat more advanced. Illustrated by Jeanne Bendick and Ted Schluenderfritz

Quiz Yourself Clever! The Periodic Table DK, 2024-10-03 Get to grips with the periodic table with this fact-packed quiz book. Learn all the ingredients that make up the universe in this quiz book for children aged 9 that breaks down the periodic table. Quiz Yourself Clever! The Periodic Table goes through all 118 elements in an accessible and kid-friendly way, showing the pure form of every element through an eye-catching image surrounded by quiz-style questions. This periodic table quiz book for children offers fast and fun learning through the use of detailed images and interesting facts. Hundreds of quick-fire questions for children to quiz themselves and challenge friends and family. Information that has been endorsed and authenticated by experts. Vivid pictures and fast facts will allow children to gather a wealth of knowledge in a fun way. Turn the page to find the answers contained in a handy dataset from its atomic number and mass to its melting point and when it was discovered. Images of how it appears in nature and how we use it give a fuller understanding of each element. More in the series. If you enjoyed Quiz Yourself Clever! The Periodic Table, then why not test yourself on other quiz books to boost your knowledge like Quiz Yourself Clever! Rocks & Minerals to get to grips with the fascinating rock and mineral forms all over the world or Animals of the World to learn about the most fascinating wildlife on our planet.

The Periodic Table Mary Griffin, 2018-12-15 The periodic table is the way scientists have organized the known elements, but up and coming scientists can learn much from the periodic table as well. Your readers will learn how to read the periodic table, including what an element's atomic number means and what the rows and columns signify. They'll gather facts about common and interesting elements and even discover how they can predict how elements might chemically react from their position in the table. Fact boxes and images add additional information to the accessible text.

The Periodic Table Tom Jackson, 2020-02-04 Which is the densest element? Which has the largest atoms? And why are some elements radioactive? From the little-known uses of gold in medicine to the development of the hydrogen bomb, this is a fresh new look at the Periodic Table. Combining cutting-edge science with fascinating facts and stunning infographics, this book looks at the extraordinary stories of discovery, amazing properties and surprising uses of each element, whether solid, liquid or gas, naturally occurring, synthesised or theoretical. From hydrogen to Oganesson, this is a fact-filled visual guide to each element, each accompanied by technical data: category, atomic number, weight, boiling point, as well as fun facts and stories about their discovery and surprising uses.

Periodic Table Cindy Blobaum, 2005 What do chocolate chip cookies, chemistry and logic have in common? They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table. This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an

element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket *The Secret Life of the Periodic Table* Dr Ben Still, Hamlyn, 2025-07-17 Every element has character be it volatile aloof gregarious or enigmatic They also have incredible stories of how they came to be how they were discovered and how their qualities have been harnessed to make everything we have in the world Dr Ben Still takes you through all the groups revealing the fascinating circumstances of their discoveries and explaining their characteristics and behaviour often delving inside the atom to explain their structure in a way easily accessible to everyone Insightful illustrations help you to understand even the most complex looking aspects of quantum physics bringing the periodic table to life Incredibly enlightening The Secret Life of the Periodic Table will entertain and inform in equal measure Take a trip through the ages of exploration enjoying the story of how Mendeleev's table came to be it is the story of one the greatest pattern seeking accomplishments of humankind Contents include Atomic physics The quantum atom Trends and patterns Alkaline earth metals Metalloids Future elements *Elements and the Periodic Table, Grades 5 - 8* Abbigy, 2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters quizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science *Periodic Table* DK, 2020-01-02 A brand new science ebook for kids joins this hugely successful mini encyclopedia series including lot of information Discover all the major elements of the periodic table arranged in chapters according to their group including alkaline earth metals lanthanides and noble gases Every important element from hydrogen via carbon oxygen and gold to oganesson that's element number 118 is presented here Each element is featured in its own catalogue style entry with a photo of the element in its raw state or in use a caption explaining important information ranging from where it's found to its main uses and a fact list presenting the key data including the date of discovery and all atomic information as found on the periodic table Plus there are stunning full page photos showing elements as you have never seen them before perhaps inside machinery that you can't normally delve into or in stunning macro photography that reveals microscopic details invisible to the human eye The style of the Pocket Eyewitness series is perfect for all children from reluctant readers who can easily digest the key points through to budding Marie Curies and Louis Pasteurs who want to know more about the most essential particles on the planet **Memorizing the Periodic Table of Chemical Elements** Darden Dickson, 2008-05-30 In the fall of 1997 I took an Introduction to Chemistry class Chem 100 at Tacoma Community College Tacoma Washington We were required to memorise at least eight or ten of the more important chemical elements I asked myself at the time if it was possible to memorise the number of protons and their

symbols for the entire periodic table That question grew to include memorisation of nuclear mass numbers The answer to both questions is yes and the process is a simple one as you will see The first part of the memorisation process is to use a blank Periodic Table This book provides one that you can duplicate for memory work You can also construct one with two sheets of college ruled paper a straight edge and a pen or pencil Building one for yourself is a very helpful aid in memorisation because it requires you to learn the number of chemical elements in each row and the number of rows Either way once you have a blank Periodic Table you are ready to begin your memory work First of all a chemical element is identified by the number of protons 1 Hydrogen has one Helium two Lithium has three etc The number of protons goes in sequential order from 1 to 112 each number representing an individual chemical element Memorisation requires associating the number of protons with the chemical element and its symbol Its physical place in the Periodic Table helps in memorisation This book provides a complete sequential listing of chemical elements by the number of protons Table 1 to aid you in the memorisation process The second part of the memorisation process once the symbols for the chemical elements and the number of protons have been recorded is to memorise the formulae and the series of black numbers shown in the Memorisation Key These will help you to easily derive the nuclear mass numbers 1 I have chosen the phrase the number of protons to replace the phrase the atomic numbers for philosophy of science reasons The word atom in Greek n the fall of 1997 I took an Introduction to Chemistry class Chem 100 at Tacoma Community College Tacoma Washington We were required to memorise at least eight or ten of the more important chemical elements I asked myself at the time if it was possible to memorise the number of protons and their symbols for the entire periodic table That question grew to include memorisation of nuclear mass numbers The answer to both questions is yes and the process is a simple one as you will see The first part of the memorisation process is to use a blank Periodic Table This book provides one that you can duplicate for memory work You can also construct one with two sheets of college ruled paper a straight edge and a pen or pencil Building one for yourself is a very helpful aid in memorisation because it requires you to learn the number of chemical elements in each row and the number of rows Either way once you have a blank Periodic Table you are ready to begin your memory work First of all a chemical element is identified by the number of protons 1 Hydrogen has one Helium two Lithium has three etc The number of protons goes in sequential order from 1 to 112 each number representing an individual chemical element Memorisation requires associating the number of protons with the chemical element and its symbol Its physical place in the Periodic Table helps in memorisation This book provides a complete sequential listing of chemical elements by the number of protons Table 1 to aid you in the memorisation process The second part of the memorisation process once the symbols for the chemical elements and the number of protons have been recorded is to memorise the formulae and the series of black numbers shown in the Memorisation Key These will help you to easily derive the nuclear mass numbers 1 I have chosen the phrase the number of protons to replace the phrase the atomic numbers for philosophy of science reasons The word atom in

Greek means indivisible or uncut All the chemical elements of the Periodic Table can be broken down into smaller parts

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