45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

(1)(14.0) + (1)(16.0) 30.0 g/mo)

Water (H₂O) gmm:

(2)(1.0) + (1)(16.0) 18.0 g/mol

Yield

45 46 | 18.0 g/mol

Yield

45 46 | 18.0 g/mol

Yield

45 46 | 18.0 g/mol

Yield

46 75 gH₂O = 6.75 gH₂O

Percent Yield Calculations Answers

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