

1. Write the group numbers in the boxes along the top of the periodic table.
2. Write the period numbers in the boxes down the side of the periodic table.

[illegible]

Elements in the same group have similar properties.

Lithium (Li) and potassium (K) are metals. They are solids at room temperature, have low melting points compared to other metals and are very reactive.

3. Name two other elements that have similar properties to lithium and potassium.

Any two from:

- sodium (Na)
- rubidium (Rb)
- caesium (Cs)
- francium (Fr)

Neon (Ne) and xenon (Xe) are non-metals. They are gases at room temperature and are unreactive.

4. Name two other elements that have similar properties to neon and xenon.

Any two from:

- helium (He)
- argon (Ar)
- krypton (Kr)
- radon (Rn)

Reinforcement Periodic Table Worksheet Answers

Chapter 1

McGraw-Hill Staff



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of the 118 and counting elements and the various ways they have been classified and organized by prominent scientists up to the present day periodic table Instruction in how to read the periodic table is accompanied by examinations of the various groups of elements their location on the table and their properties and practical uses This text strongly supports Common Core Standards for the reading of scientific and technical texts and accounts and furnishes ample opportunities to summarize cite evidence and analyze connections between ideas individuals and events **Reinforcement Worksheets, Teacher Edition, for Use with Glencoe Physical Science** Aron Thompson, Glencoe/McGraw-Hill, 1999-01 **Mastering the Periodic Table** Linda Trombley, 1988

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