

1. _____ **Arrhenius bases** _____, tend to release hydroxide ions (OH^-) in solution.
2. _____ **Brønsted-Lowry acids** _____ are substances that donate a proton.
3. _____ **Lewis acids** _____ are substances that can accept a pair of electrons.
4. The _____ **pH** _____ scale describes the hydroxide ion concentration.
5. Acidic solutions have a high concentration of _____ **H_3O^+** _____ ions.
6. Solutions that have a pH above 7 are considered _____ **basic** _____.
7. A(n) _____ **amphoteric** _____ substance can act as both a Brønsted-Lowry acid and a Brønsted-Lowry base.
8. _____ **Buffers** _____ are solutions that resist pH changes.

Instructions: For each of the following questions, select the letter that best answers each question based on class notes and discussion. (1.5 points each)

- _____ 9. Which of the following is a property of acids?
- A. Have a bitter taste
 - B. Tends to turn blue litmus paper red**
 - C. Tends to turn red litmus paper blue
 - D. Have a slippery feeling
- _____ 10. What is neutralization reactions in aqueous solutions produce?
- A. A conjugate acid and conjugate base
 - B. A weak acid and a strong base
 - C. A salt and water**
 - D. A salt and a conjugate acid
- _____ 11. The scientist(s) who classified acids and bases by the ions they produce in solution were
- A. Sarrusens
 - B. Lewis
 - C. Brønsted and Lowry
 - D. Arrhenius**
- _____ 12. Using the Brønsted-Lowry model, strong acids

Reading Essentials Acids Bases And Salts Answer Key

M Woodhall



Reading Essentials Acids Bases And Salts Answer Key:

Egan's Fundamentals of Respiratory Care - E-Book Robert M. Kacmarek, James K. Stoller, Albert J. Heuer, 2012-04-19 Updated content aligned with the 2009 NBRC CRT Summary Content Outline ensures the text is both current and clinically accurate Expanded use of the NBRC Exam Matrix Correlation Chart throughout all Evolve online resources makes test preparation easier

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and the process of science A supporting theme throughout the text continually emphasizes that chemistry is everywhere

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