

10.4 Electrophilic Reactions

General mechanism An electrophile attacks, adding across the double bond.

Example: Bromination

Chloroethene reacts with bromine, as shown in the scheme below and steps:

1. A curly arrow represents the movement of an electron pair from the $\text{C}=\text{C}$ double bond to the $\text{Br}-\text{Br}$ bond.
2. The electrophile attacks the double bond of the $\text{C}=\text{C}$ and breaks the $\text{Br}-\text{Br}$ bond, forming a curly arrow.

Step 1: The curly arrow represents the movement of an electron pair from the $\text{C}=\text{C}$ double bond to the $\text{Br}-\text{Br}$ bond.

Step 2: The curly arrow represents the movement of an electron pair from the $\text{C}=\text{C}$ double bond to the $\text{Br}-\text{Br}$ bond.

Step 3: The curly arrow represents the movement of an electron pair from the $\text{C}=\text{C}$ double bond to the $\text{Br}-\text{Br}$ bond.

When writing a mechanism, it is essential to show curly arrows.

Practice: Electrophilic Reactions

1. In the following mechanism, a curly arrow is shown. Complete the mechanism by drawing curly arrows to show the movement of electron pairs.
2. Write down the name of the electrophile in the reaction.
3. Write down the name of the nucleophile in the reaction.
4. Write down the name of the product in the reaction.
5. Write down the name of the reagent in the reaction.
6. Write down the name of the solvent in the reaction.
7. Write down the name of the catalyst in the reaction.
8. Write down the name of the reaction in the reaction.

Questions

1. Write down the name of the electrophile in the reaction.

Reading And Study Workbook 19 4 Neutralization Reactions

Jin-Ying Zhang



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Table of Contents Reading And Study Workbook 19 4 Neutralization Reactions

1. Understanding the eBook Reading And Study Workbook 19 4 Neutralization Reactions
 - The Rise of Digital Reading Reading And Study Workbook 19 4 Neutralization Reactions
 - Advantages of eBooks Over Traditional Books
2. Identifying Reading And Study Workbook 19 4 Neutralization Reactions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reading And Study Workbook 19 4 Neutralization Reactions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Reading And Study Workbook 19 4 Neutralization Reactions

- Personalized Recommendations
- Reading And Study Workbook 19 4 Neutralization Reactions User Reviews and Ratings
- Reading And Study Workbook 19 4 Neutralization Reactions and Bestseller Lists
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 - Reading And Study Workbook 19 4 Neutralization Reactions Public Domain eBooks
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 - ePub, PDF, MOBI, and More
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 - Reading And Study Workbook 19 4 Neutralization Reactions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reading And Study Workbook 19 4 Neutralization Reactions
 - Highlighting and Note-Taking Reading And Study Workbook 19 4 Neutralization Reactions
 - Interactive Elements Reading And Study Workbook 19 4 Neutralization Reactions
- 8. Staying Engaged with Reading And Study Workbook 19 4 Neutralization Reactions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reading And Study Workbook 19 4 Neutralization Reactions
- 9. Balancing eBooks and Physical Books Reading And Study Workbook 19 4 Neutralization Reactions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reading And Study Workbook 19 4 Neutralization Reactions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Reading And Study Workbook 19 4 Neutralization Reactions
 - Setting Reading Goals Reading And Study Workbook 19 4 Neutralization Reactions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Reading And Study Workbook 19 4 Neutralization Reactions

- Fact-Checking eBook Content of Reading And Study Workbook 19 4 Neutralization Reactions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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