

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

# NOVA: Hunting the Elements

## Part 1: Basic Chemistry

### Periodic Table of Elements

Use the periodic table below to help you answer some questions covered in the documentary.

|                          |  |                     |  |                       |  |                     |  |                      |  |                       |  |                       |  |                     |  |                        |  |                       |  |                      |  |                        |  |                     |  |                      |  |                     |  |                      |  |                    |  |                    |  |                   |  |
|--------------------------|--|---------------------|--|-----------------------|--|---------------------|--|----------------------|--|-----------------------|--|-----------------------|--|---------------------|--|------------------------|--|-----------------------|--|----------------------|--|------------------------|--|---------------------|--|----------------------|--|---------------------|--|----------------------|--|--------------------|--|--------------------|--|-------------------|--|
| 1<br>H<br>1.00794        |  |                     |  |                       |  |                     |  |                      |  |                       |  |                       |  |                     |  |                        |  | 2<br>He<br>4.002602   |  |                      |  |                        |  |                     |  |                      |  |                     |  |                      |  |                    |  |                    |  |                   |  |
| 3<br>Li<br>6.941         |  | 4<br>Be<br>9.012182 |  |                       |  |                     |  |                      |  |                       |  |                       |  |                     |  | 5<br>B<br>10.811       |  | 6<br>C<br>12.011      |  | 7<br>N<br>14.00643   |  | 8<br>O<br>15.999       |  | 9<br>F<br>18.998403 |  | 10<br>Ne<br>20.1797  |  |                     |  |                      |  |                    |  |                    |  |                   |  |
| 11<br>Na<br>22.98976928  |  | 12<br>Mg<br>24.304  |  |                       |  |                     |  |                      |  |                       |  |                       |  |                     |  | 13<br>Al<br>26.9815386 |  | 14<br>Si<br>28.08558  |  | 15<br>P<br>30.973762 |  | 16<br>S<br>32.06       |  | 17<br>Cl<br>35.453  |  | 18<br>Ar<br>39.948   |  |                     |  |                      |  |                    |  |                    |  |                   |  |
| 19<br>K<br>39.0983       |  | 20<br>Ca<br>40.078  |  | 21<br>Sc<br>44.955912 |  | 22<br>Ti<br>47.88   |  | 23<br>V<br>50.9415   |  | 24<br>Cr<br>51.9961   |  | 25<br>Mn<br>54.938045 |  | 26<br>Fe<br>55.845  |  | 27<br>Co<br>58.933195  |  | 28<br>Ni<br>58.6934   |  | 29<br>Cu<br>63.546   |  | 30<br>Zn<br>65.38      |  | 31<br>Ga<br>69.723  |  | 32<br>Ge<br>72.64    |  | 33<br>As<br>74.9216 |  | 34<br>Se<br>78.96    |  | 35<br>Br<br>79.904 |  | 36<br>Kr<br>83.80  |  |                   |  |
| 37<br>Rb<br>85.4678      |  | 38<br>Sr<br>87.62   |  | 39<br>Y<br>88.90584   |  | 40<br>Zr<br>91.224  |  | 41<br>Nb<br>92.90638 |  | 42<br>Mo<br>95.94     |  | 43<br>Tc<br>98        |  | 44<br>Ru<br>101.07  |  | 45<br>Rh<br>102.9055   |  | 46<br>Pd<br>106.90558 |  | 47<br>Ag<br>107.8682 |  | 48<br>Cd<br>112.411    |  | 49<br>In<br>114.818 |  | 50<br>Sn<br>118.710  |  | 51<br>Sb<br>121.757 |  | 52<br>Te<br>127.6    |  | 53<br>I<br>126.905 |  | 54<br>Xe<br>131.29 |  |                   |  |
| 55<br>Cs<br>132.90545196 |  | 56<br>Ba<br>137.327 |  | 57-70<br>*            |  | 71<br>Lu<br>174.967 |  | 72<br>Hf<br>178.49   |  | 73<br>Ta<br>180.94788 |  | 74<br>W<br>183.84     |  | 75<br>Re<br>186.207 |  | 76<br>Os<br>190.23     |  | 77<br>Ir<br>192.222   |  | 78<br>Pt<br>195.084  |  | 79<br>Au<br>196.966569 |  | 80<br>Hg<br>200.59  |  | 81<br>Tl<br>204.3833 |  | 82<br>Pb<br>207.2   |  | 83<br>Bi<br>208.9804 |  | 84<br>Po<br>[209]  |  | 85<br>At<br>[210]  |  | 86<br>Rn<br>[222] |  |
| 87<br>Fr<br>[223]        |  | 88<br>Ra<br>[226]   |  | 89-102<br>*-*         |  | 103<br>Lr<br>[260]  |  | 104<br>Rf<br>[261]   |  | 105<br>Db<br>[262]    |  | 106<br>Sg<br>[266]    |  | 107<br>Bh<br>[264]  |  | 108<br>Hs<br>[277]     |  | 109<br>Mt<br>[268]    |  | 110<br>Uun<br>[289]  |  | 111<br>Uuu<br>[288]    |  | 112<br>Uub<br>[285] |  | 113<br>Uut<br>[284]  |  | 114<br>Uuq<br>[289] |  |                      |  |                    |  |                    |  |                   |  |

\* Lanthanide series

|                       |                    |                       |                     |                   |                    |                     |                     |                       |                      |                       |                     |                       |                       |
|-----------------------|--------------------|-----------------------|---------------------|-------------------|--------------------|---------------------|---------------------|-----------------------|----------------------|-----------------------|---------------------|-----------------------|-----------------------|
| 57<br>La<br>138.90547 | 58<br>Ce<br>140.12 | 59<br>Pr<br>140.90766 | 60<br>Nd<br>144.242 | 61<br>Pm<br>[145] | 62<br>Sm<br>150.36 | 63<br>Eu<br>151.964 | 64<br>Gd<br>157.254 | 65<br>Tb<br>158.92532 | 66<br>Dy<br>162.5001 | 67<br>Ho<br>164.93032 | 68<br>Er<br>167.259 | 69<br>Tm<br>168.93032 | 70<br>Yb<br>173.05468 |
|-----------------------|--------------------|-----------------------|---------------------|-------------------|--------------------|---------------------|---------------------|-----------------------|----------------------|-----------------------|---------------------|-----------------------|-----------------------|

\*\* Actinide series

|                   |                      |                       |                      |                   |                   |                   |                   |                   |                   |                   |                    |                    |                    |
|-------------------|----------------------|-----------------------|----------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|--------------------|--------------------|--------------------|
| 89<br>Ac<br>[227] | 90<br>Th<br>232.0377 | 91<br>Pa<br>231.03688 | 92<br>U<br>238.02891 | 93<br>Np<br>[237] | 94<br>Pu<br>[244] | 95<br>Am<br>[243] | 96<br>Cm<br>[247] | 97<br>Bk<br>[247] | 98<br>Cf<br>[251] | 99<br>Es<br>[252] | 100<br>Fm<br>[257] | 101<br>Md<br>[258] | 102<br>No<br>[259] |
|-------------------|----------------------|-----------------------|----------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|--------------------|--------------------|--------------------|

### Gold - Au

- Write the number of subatomic particles in gold:
  - Protons: **79**
  - Neutrons: **118**
  - Electrons: **79**
- Give one property of gold.  
**shiny, malleable (soft), high density, electrically conductive**
- How much gold is in one ton of the mined rock?  
**1 ounce**

# Nova Hunting The Elements Worksheet Answers

**M Planty**



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