

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ relationship _____ equated _____

2. $40g = 40r + 4 \cdot 20g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ _____ from the _____ reaction in the _____

Ex. Problem

When aluminum chloride reacts with elemental sulfur, aluminum chloride is produced



moles of

$2AlCl_3$	$3S$	$2Al_2S_3$
$2AlCl_3$	$3S$	$2Al_2S_3$
$2AlCl_3$	$3S$	$2Al_2S_3$

2. How many moles of Al_2S_3 will be produced?

2. $2AlCl_3 + 3S \rightarrow 2Al_2S_3$

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Didier Musso



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Enjoying the Melody of Phrase: An Emotional Symphony within **Note Taking Guide Episode 801 Answers Gpb**

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