1. Particles

- If we could divide any substance down into smaller and smaller pieces we could see it is made of tiny particles.
- These particles are so small they could not be seen-using a microscope.
- We can represent these particles in models using saftenes.



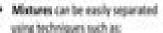
4. Changes in states of matter



 A solid melts when it is heated because the particles are gaining energy. This energy is used to break the forces of attraction between the molecules. The more energy a particle has the faster it can move.

6. Elements and mixtures

 This diagram shows a mixture of elements.



- Filtration
- Dividitation
- Chromatography





2. States of matter

. Three states of matter are solld. Rould and gas.

Solid	Liquid	Gas
900	43	•

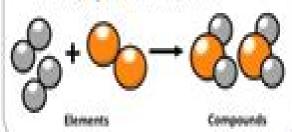


KS3 Spine

All matter in the universe is made of small particles

T. Compounds

 Atoms of different elements can combine together to form a very large number of compounds.



3. Oifferences in states of matter

- In solids the particles are closely spaced and vibrating.
 In liquids there is random motion but the particles are in contact.
- In a gas there is random motion and the particles are widely spaced.
- Liquids and gooss flow and completely fill their container, Gases can be compressed.

5. Atoms and elements

- Atoms are the basic building blocks of all living and non-living things anywhere in the universe.
- There are just over \$00 different types of atom.
- Substances made of only one kind of atom are called elements.
- All the known elements are listed in the periodic table of elements.

8. Reactions and properties

- A chemical reaction involves the rearrangement of atoms to form new substances, while the total number of atoms stays the same (see image above).
- The properties of different materials can be explained in terms of the behaviour of the atoms and groups of atoms of which they are made.

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