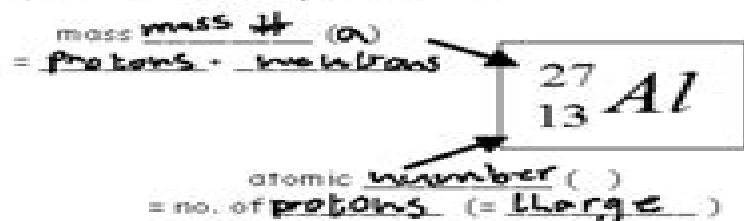


**Nucleons:**

- are particles occupying the nucleus
- consist of + charged protons and neutral charge neutrons
- have almost 2000 times the mass of electrons
- are made up of quark and leptons

**Nuclear Notation:**



${}_{13}^{27}\text{Al}$  has 13 protons and 14 neutrons, for a total of 27 nucleons.

${}_{13}^{28}\text{Al}$  has 13 protons and 15 neutrons, for a total of 28 nucleons.

**isotopes:** atoms of same element with different numbers of neutrons (different masses)

**Forces Acting on Nucleons:**

strong forces -

- forces of attraction between nucleons
- are independent of the charges of the nucleon
- are short range (exist only between closest neighbors)

electrical force -

- force of repulsion between + charged protons
- are long range

**When are nuclei unstable?**

- large nuclei ( $Z > 82$ ) - electrical forces of repulsion are greater than strong forces of attraction
- wrong neutron : proton ratio

**A radioactive isotope:**

- has an unstable nucleus
- spontaneously emits a particle and charge into another element

# Note Taking Guide Episode 4answers Chemistry

**D Kirk**



## **Note Taking Guide Episode 4answers Chemistry:**

## Decoding **Note Taking Guide Episode 4answers Chemistry**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Note Taking Guide Episode 4answers Chemistry**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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