
NIMS 100 Exam 2024 Questions & Answers 100% Correct!

ICS - ANSWERSincident command system

command - ANSWERSthe act of directing, ordering, or controlling by virtue of explicit statutory, regulatory, or delegated authority

chain of command - ANSWERSorderly line of authority within the ranks of the incident management organization

transfer of command - ANSWERSthe process of moving the responsibility for incident command from one Incident Commander to another

unity of command - ANSWERSevery individual is accountable to only one designated supervisor, to whom they report at the scene of an incident

incident objectives - ANSWERSgoals that are used to ensure that everyone within the ICS understands what needs to be accomplished

incident objective priorities - ANSWERS1. life safety
2. incident stabilization
3. property preservation

Incident Action Plan (IAP) - ANSWERSplan for incident that includes the incident objectives, the activities that need to be completed, a specified timeframe, assigned responsibilities, and identified needed resources

operational period - ANSWERSspecified timeframe within an incident action plan

resources - ANSWERSpersonnel, supplies, and equipment

incident command post - ANSWERSthe location from which the Incident Commander oversees all incident operations

staging area - ANSWERSa temporary location where personnel and equipment are gathered while waiting to be assigned

incident base - ANSWERSthe location from which primary logistics and administrative functions are coordinated and administered

Nims 100 Test Answers With Questions

**Yew Soon Ong, Liang Feng, Huajin
Tang, Will Neil Browne**



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web solution 1 the two half reactions balanced as if in acidic solution $2\text{CN}^- + \text{Au} \rightarrow \text{AuCN}_2^- + 2\text{e}^-$ $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2 + 2\text{e}^-$ make electrons equal convert to basic solution $4\text{CN}^- + 2\text{Au} \rightarrow 2\text{AuCN}_2^- + 2\text{e}^-$ multiplied by a factor of 2 $2\text{H}_2\text{O}_2 + \text{O}_2 + 2\text{H}_2\text{O} + 3$ the final answer

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web practice exercises balance the following equations of redox reactions assign oxidation numbers to all elements in the reaction separate the redox reaction into two half reactions balance the atoms in each half reaction add the two half reactions together and cancel out common terms

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web aug 29 2023 some points to remember when balancing redox reactions the equation is separated into two half equations one for oxidation and one for reduction the equation is balanced by adjusting coefficients and adding H^+ H_2O and e^- in this order balance the atoms in the equation apart from oxygen and hydrogen

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{HNO}_2 + \text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} + 2\text{NO}$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 a $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 b NO_2 and $\text{C}_6\text{H}_8\text{O}_6$ c $\text{C}_6\text{H}_8\text{O}_6$ and HNO_2 d HNO_2 and $\text{C}_6\text{H}_8\text{O}_6$