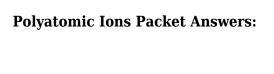
# List of Common Polyatomic Ions

Cations: Symbol	Name	Charge	Symbol	Name	Charge
NH4°	Ammonium	+1	$Hg_2^{+2}$	Dimercury (I)	+2
Anions:					
Symbol	<u>Name</u>	Charge	Symbol	Name	Charge
$C_2H_3O_2$	Acetate	- 1	CO <sub>3</sub> -2	Carbonate	-2
HCO <sub>3</sub>	Bicarbonate	- 1	$CrO_4^{-2}$	Chromate	-2
BrO <sub>3</sub>	Bromate	- 1	$Cr_2O_7^{-2}$	Dichromate	-2
BrO <sub>2</sub>	Bromite	- 1	$C_2O_4^{-2}$	Oxalate	-2
ClO <sub>4</sub>	Perchlorate	- 1	$O_2^{-2}$	Peroxide	-2
ClO <sub>3</sub>	Chlorate	- 1	SO <sub>4</sub> -2	Sulfate	-2
CIO <sub>2</sub>	Chlorite	- 1	SO <sub>3</sub> -2	Sulfite	-2
CIO	Hypochlorite	- 1	S <sub>2</sub> O <sub>3</sub> -2	Thiosulfate	-2
CN	Cyanide	- 1	SiO <sub>3</sub> 2	Silicate	-2
SCN	Thiocyanate	- 1	$MoO_4^{-2}$	Molybdate	-2
OH	Hydroxide	- 1	$WO_4^{-2}$	Tungstate	-2
NO <sub>3</sub>	Nitrate	- 1	$BO_3^{-3}$	Borate	-3
NO2	Nitrite	-1	$PO_4^{-3}$	Phosphate	-3
MnO <sub>4</sub>	Permanganate	- 1	$PO_{3}^{-3}$	Phosphite	-3
HSO.	Bisulfate	- 1	AsO <sub>4</sub> -3	Arsenate	-3
HSO <sub>3</sub>	Bisulfite	- 1	AsO <sub>3</sub> -3	Arsenite	-3
IO <sub>3</sub>	Iodate	- 1			
IO4	Periodate	- 1			
C <sub>6</sub> H <sub>5</sub> COO	Benzoate	-1			

Symbol	Name	Charge
$H_{2}PO_{4}^{-1}$	Dihydrogen phosphate	- 1
$HPO_4^{-2}$	Hydrogen phosphate	-2
Fe(CN) <sub>6</sub> -3	Hexacyannoferrate (III)	-3
Fe(CN) <sub>6</sub> -4	Hexacyannoferrate (II)	-4

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**CL Gary** 



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# **Table of Contents Polyatomic Ions Packet Answers**

- 1. Understanding the eBook Polyatomic Ions Packet Answers
  - The Rise of Digital Reading Polyatomic Ions Packet Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Polyatomic Ions Packet Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Polyatomic Ions Packet Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Polyatomic Ions Packet Answers
  - Personalized Recommendations
  - Polyatomic Ions Packet Answers User Reviews and Ratings
  - Polyatomic Ions Packet Answers and Bestseller Lists
- 5. Accessing Polyatomic Ions Packet Answers Free and Paid eBooks

- Polyatomic Ions Packet Answers Public Domain eBooks
- Polyatomic Ions Packet Answers eBook Subscription Services
- Polyatomic Ions Packet Answers Budget-Friendly Options
- 6. Navigating Polyatomic Ions Packet Answers eBook Formats
  - o ePub, PDF, MOBI, and More
  - Polyatomic Ions Packet Answers Compatibility with Devices
  - Polyatomic Ions Packet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Polyatomic Ions Packet Answers
  - Highlighting and Note-Taking Polyatomic Ions Packet Answers
  - Interactive Elements Polyatomic Ions Packet Answers
- 8. Staying Engaged with Polyatomic Ions Packet Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Polyatomic Ions Packet Answers
- 9. Balancing eBooks and Physical Books Polyatomic Ions Packet Answers
  - Benefits of a Digital Library
  - $\circ\,$  Creating a Diverse Reading Collection Polyatomic Ions Packet Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Polyatomic Ions Packet Answers
  - Setting Reading Goals Polyatomic Ions Packet Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Polyatomic Ions Packet Answers
  - Fact-Checking eBook Content of Polyatomic Ions Packet Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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