### Polyatomic Ions Worksheet Answer Key

#### Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

sulfate	SO?	CO <sub>3</sub> 2-	carbonate	
nitrite	NO;	MnO <sub>3</sub> 1-	manganate	
perphosphate	PO3-	SO <sub>5</sub> <sup>2-</sup>	persulfate	
hypoiodite	10	BrO <sub>2</sub> 1-	bromite	
chlorite	CIO,	OO42-	percarbonate	
phosphite	PO3-	POs <sup>3-</sup>	perphosphate	
percarbonate	CO <sub>4</sub>	CIO <sup>1-</sup>	hypochlorite	
bromate	BrO;	1O25-	iodite	
hyposulfite	SO <sub>2</sub> -	PO43-	phosphate	
permanganate	MnO.	NO <sub>2</sub> 1-	nitrite	
carbonite	CO2	SO <sub>4</sub> 2-	sulfate	

2. Name or write the formula for the following Type I polyatomic ionic compounds

beryllium hydroxide	Be(OH) <sub>2</sub>	Ba(IO <sub>3</sub> ) <sub>2</sub>	barium iodate
sodium nitrite	NaNO <sub>2</sub>	Ga(CNO) <sub>3</sub>	gallium cyanate
ammonium chloride	NH,CI	Ag <sub>2</sub> SO <sub>3</sub>	silver sulfite
calcium bisulfate	Ca(HSO <sub>4</sub> ) <sub>2</sub>	MgCO	magnesium hyp-ocarbonite
rubidium perchlorate	RECIO.	NH <sub>3</sub> NO <sub>2</sub>	ammonium nitrite
strontium sulfite	SrSO,	AI(C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> ) <sub>3</sub>	aluminum acetate
aluminum acetate	AI(C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> ) <sub>3</sub>	SrSO <sub>5</sub>	strontium persulfate
ammonium nitrate	NH,NO,	RbClO <sub>2</sub>	rubidium chlorite
magnesium hypocarbonite	MgCO	Ca(HSO <sub>4</sub> ) <sub>2</sub>	calcium bisulfate
silver hyposulfite	Ag <sub>2</sub> SO <sub>2</sub>	NH <sub>4</sub> CI	ammonium chloride

# **Polyatomic Ions Worksheet Answer Key Unit 3**

**Daniel F McAuley** 

**Polyatomic Ions Worksheet Answer Key Unit 3:** 

Whispering the Techniques of Language: An Psychological Journey through **Polyatomic Ions Worksheet Answer Key Unit 3** 

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