

## Exam 1 Review

### Topics:

1. General Chemistry
2. Molecular Representations
3. Nomenclature
4. Alkanes/Cycloalkanes
5. Stereoisomers
6. Chemical reactivity/mechanism
7. SN/E Reactions

### General Chemistry Questions

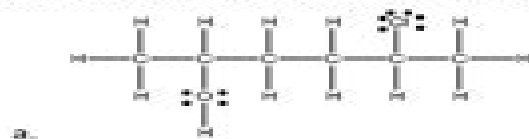
1. Identify all of the intermolecular forces present in each molecule:
  - a.  $\text{CH}_3\text{OH}$
  - b.  $\text{CH}_3\text{COCH}_2\text{CH}_3$
  - c.  $\text{CH}_3\text{C}(\text{CH}_3)_2$
2. Fill out the blanks in the chart (VSEPR Theory)

| Hybridization | Geometry    | # of Bonds | # of Lone Pairs | Bond Angle |
|---------------|-------------|------------|-----------------|------------|
| $\text{Sp}^3$ | Tetrahedral | 3          | 0               |            |
|               | Bent        |            | 1               | 109.5      |
| $\text{Sp}^2$ |             | 3          | 0               |            |
|               |             | 2          | 1               | 120        |
|               |             |            | 0               | 180        |

3. Determine if the following molecules have a dipole moment
  - a.  $\text{CH}_3\text{CH}_3$
  - b.  $\text{CH}_3\text{CH}_2\text{CH}(\text{NH}_2)\text{CH}_3$
  - c.  $\text{CH}_2\text{Cl}_2$
  - d.  $\text{H}_3\text{O}^+$
  - e.  $\text{HCO}_2^-$

### Molecular Representations Questions:

1. Write the condensed name for the following structures:



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**H Kauffman**



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## **Table of Contents Sample Questions On Chemistry 1exam**

1. Understanding the eBook Sample Questions On Chemistry 1exam
  - The Rise of Digital Reading Sample Questions On Chemistry 1exam
  - Advantages of eBooks Over Traditional Books
2. Identifying Sample Questions On Chemistry 1exam
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Sample Questions On Chemistry 1exam
  - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Questions On Chemistry 1exam
  - Personalized Recommendations
  - Sample Questions On Chemistry 1exam User Reviews and Ratings
  - Sample Questions On Chemistry 1exam and Bestseller Lists
5. Accessing Sample Questions On Chemistry 1exam Free and Paid eBooks
  - Sample Questions On Chemistry 1exam Public Domain eBooks
  - Sample Questions On Chemistry 1exam eBook Subscription Services
  - Sample Questions On Chemistry 1exam Budget-Friendly Options
6. Navigating Sample Questions On Chemistry 1exam eBook Formats

- ePub, PDF, MOBI, and More
  - Sample Questions On Chemistry 1exam Compatibility with Devices
  - Sample Questions On Chemistry 1exam Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Sample Questions On Chemistry 1exam
    - Highlighting and Note-Taking Sample Questions On Chemistry 1exam
    - Interactive Elements Sample Questions On Chemistry 1exam
  8. Staying Engaged with Sample Questions On Chemistry 1exam
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Sample Questions On Chemistry 1exam
  9. Balancing eBooks and Physical Books Sample Questions On Chemistry 1exam
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Sample Questions On Chemistry 1exam
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Sample Questions On Chemistry 1exam
    - Setting Reading Goals Sample Questions On Chemistry 1exam
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Sample Questions On Chemistry 1exam
    - Fact-Checking eBook Content of Sample Questions On Chemistry 1exam
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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