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b) propyne c) ethyne a) Butyne d) pentyne

2) The loss of _____ and ____ atoms from the double-bonded carbon atoms leads to alkyne formation.

 a) Oxygen, halogen
 b) oxygen, hydrogen
 c) hydrogen, d) carbon, halogen halogen

What is the reagent used and appropriate temperature required to prepare ethene from ethyl alcohol?

a) Diluted H₂SO₄, 160°C

b) Concentrated H₂SO₄, 170°C

c) Diluted HCl, 170°C

d) Concentrated HCl, 160°C

4) _____ on heating with alcoholic potash loses one molecule of hydrogen halide to give alkene.

a) Halogen

b) Haloalkanes

c) Haloalkynes

d) None of the above

5) The IUPAC name of the compound H2C=CH-CH(CH3)2:

a) 2-Vinylpropene

b) 1,1-dimethyl-2-peopene

c) 3-Methyl-1-butene

d) 1-Isopropyl ethylene

6) Ethylbenzene reacts with Cl₂ in sunlight to give:

a) I-Chloro-1-phenylethane

b) 1-Chloro-2-phenylethane

c) 4-Chloroethylbenzene

d) 3-Chloroethylbenzene

7) Benzene Can be :

a) Discharge the color of KMNO₄

b) Discharge the color of bromine water

c) Polymerize

d) Undergo substitution reactions

8) Which of the following reactions will yield 2, 2 -dibromopropane?

a) HC == CH + 2HBr -->

b) H₃C — ⊆ = CH₂ + HB_F — ►

c) H₃C — C == CHBr + 2HBr — ► d) H₃C — C == CHBr + HBr — ►

Give the structure of reactants :

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