

Name _____

Chemical Reactions Quiz (Quiz ID HMLJW2I)

Score

- The law that states that atoms do not appear or disappear in chemical reactions, is the Law of
- (A) Conservation of Energy
- (B) Atom Stability
- C Conservation of Matter
- (D) Balancing Equations
- In a chemical reaction you begin with 18 atoms. When the reaction is over you would expect to have ______ atoms remaining.
- (A) less than 18
- (B) 18
- (C) more than 18
- (D) Not enough info to determine this
- 3. Which statement correctly supports the argument that the illustrated chemical equation shown violates the law of conservation of matter?

$$H_2 + O_2 \rightarrow 2H_2O$$

- A There are more oxygen atoms on the left side of the arrow then on the right side
- (B) There are more hydrogen atoms on the left side of the arrow then on the right side.
- C There are more oxygen atoms on the right side of the arrow then on the left side.
- There are more hydrogen atoms on the right side of the arrow then on the left side
- 4. Which is the most accurate description of the changes that take place in atoms when they form chemical bonds?
- (A) Atoms gain, share, or lose protons
- B Atoms gain, share, or lose neutrons
- Atoms gain, share, or lose electrons
- (D) Atoms gain, share, or lose positrons

Quiz Section Describing Chemical Reactions

Ying-Ying Zheng

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