

Assignment 2

Heat

Section One: Defining Heat

Write the letter of the correct answer in the space provided.

- _____ 1. What state is achieved as the energy transferred between objects with different temperatures?
 - a. constant energy
 - b. zero
 - c. zero
 - d. thermal equilibrium
- _____ 2. What would happen if energy is or has transferred between two bodies in physical contact?
 - a. The two bodies reach the same volume.
 - b. The two bodies reach the same temperature.
 - c. The two bodies reach the same different volume.
 - d. The two bodies reach the same different temperature.
- _____ 3. What comes to the equilibrium is achieved at the temperature when energy is transferred to the substance as heat?
 - a. The average kinetic energy of the particles increases.
 - b. The average kinetic energy of the particles remains constant.
 - c. The average kinetic energy of the particles decreases.
 - d. The average kinetic energy of the particles reaches equilibrium.
- _____ 4. If energy is transferred spontaneously, as heat from a substance at T_1 to a substance at T_2 , which of the following statements must be true?
 - a. $T_1 > T_2$
 - b. $T_1 = T_2$
 - c. $T_1 < T_2$
 - d. More information is needed.
- _____ 5. A substance with a temperature of 10°C is submerged in a beaker of water with a temperature of 20°C . The substance of substance with a temperature of T_1 is submerged in an identical beaker of water with a temperature of T_2 . If the amount of energy transferred as heat between the substance and the water in the beaker for both cases, which of the following statements is true?
 - a. $T_1 = 10^\circ\text{C}$, $T_2 = 10^\circ\text{C}$
 - b. $T_1 = 10^\circ\text{C}$, $T_2 = 20^\circ\text{C}$
 - c. $T_1 = 20^\circ\text{C}$, $T_2 = 20^\circ\text{C}$
 - d. $T_1 = 7^\circ\text{C}$, $T_2 = 19^\circ\text{C}$

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