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web in a calorimeter the white solid  $CaSO_4$  forms and the temperature of the mixture increases to 28.1 °C assuming that the calorimeter absorbs only a negligible quantity of heat and that the specific heat capacity of the solution is 4.18 J/g·°C and that the density of the final solution is 1.0 g/mL calculate the enthalpy change per mole

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