

# CHEMISTRY



**Properties of Matter**

Matter is anything that has mass and takes up space. It can be solid, liquid, or gas. Matter is made of tiny particles called atoms and molecules.

**States of Matter**

- Solid: Particles are closely packed and have fixed positions. They vibrate but do not move from their places.
- Liquid: Particles are close together but can move past each other. They have more freedom of movement than in a solid.
- Gas: Particles are far apart and move freely in all directions. They have the most freedom of movement.

**Changes of State**

Changes of state occur when heat is added or removed from a substance. The three main changes are melting, boiling, and condensation.

**Atomic Structure**

Atoms are the smallest particles of an element that can take part in a chemical reaction. They are made of three subatomic particles: electrons, protons, and neutrons.

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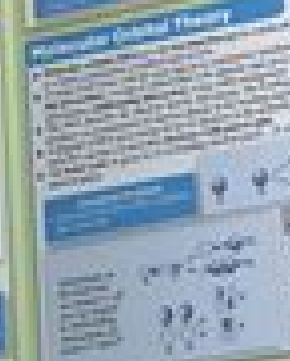
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