

JEE-Mains-25-01-2023 (Memory Based) [Shift-2] Chemistry

SECTION – A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer :

- If $[H^+]$ in concentration is increased by a factor of 1000. Then pH?
 - Decreased by 3
 - Increased by 3
 - There is no change in pH
 - Decreased by 1

Answer (1)

Sol. If $[H^+]$ is increased by 1000 times then pH will be decreased by 3.

- Arrange the following elements in increasing order of metallic character
Si, K, Mg and Be
 - Si < Mg < Be < K
 - Be < Mg < Si < K
 - Si < Be < Mg < K
 - K < Mg < Si < Be

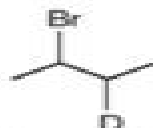
Answer (3)

Sol. Based on the electronegativity of the given elements, the correct increasing order of metallic character is
Si < Be < Mg < K

- Which of the following has two chiral centres
 - 2- Bromo - 3- deuto butane
 - 1- Bromo - 2- deuto butane
 - 1- Bromo - 3- deuto butane
 - 1- Bromo - 4- deuto butane

Answer (1)

Sol.



2 - Bromo - 3 deuto butane has two chiral centres.

- A** : Carbon form two oxides CO and CO_2 , where CO is neutral while CO_2 is acidic.

R : CO_2 will combine with water to give carbonic acid and CO is soluble in water

 - [A] and [R] both are correct and [R] is correct explanation of [A]
 - [A] and [R] both are correct and [R] is not correct explanation of [A]
 - [A] is correct while [R] is false
 - [A] is false while [R] is correct

Answer (2)

Sol. CO_2 will form carbonic acid with water and it is acidic in nature, while CO is neutral but there is no relation of neutrality with solubility.

- Which of the following element is the weakest reducing agent in aqueous solution.
 - Na
 - K
 - Li
 - Rb

Answer (1)

Sol. As per the standard reduction potential values, Na is the weakest reducing agent.

- Match List-I with List-II

(List-I)

Amine

List-II

pK_b (Aqueous medium)

- | | |
|--|---------|
| (a) Aniline | 1. 9.0 |
| (b) Ethanamine | 2. 3.29 |
| (c) N-ethylethanamine | 3. 3.25 |
| (d) N, N-diethylethanamine | 4. 3.0 |
| (1) $a \rightarrow 1, b \rightarrow 2, c \rightarrow 4, d \rightarrow 3$ | |
| (2) $a \rightarrow 1, b \rightarrow 4, c \rightarrow 3, d \rightarrow 2$ | |
| (3) $a \rightarrow 1, b \rightarrow 2, c \rightarrow 3, d \rightarrow 4$ | |
| (4) $a \rightarrow 2, b \rightarrow 3, c \rightarrow 4, d \rightarrow 1$ | |

Answer (1)

Sol. The order of basicity is:

$c > d > b > a$

$\therefore pK_b$ order is : $\rightarrow c < d < b < a$

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