

Conduction and Convection

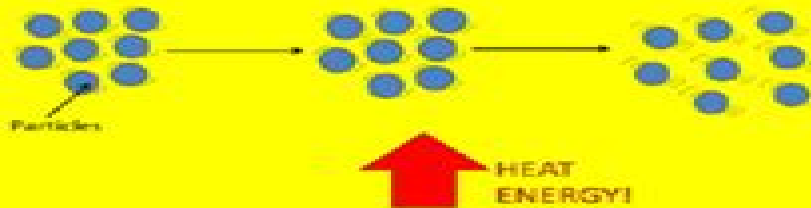
Convection Currents

Definition

Convection is the transfer of heat in liquids and gases.

The hotter the liquid/gas the particles move faster and spread out. This means the gas/liquid becomes less dense.

The less dense gas/liquid rises and the more dense gas/liquid sinks.



Because the particles have to rise and fall for convection to occur it can only happen in liquids and gases, where the particles are free to move.



Model Question (3)

Explain how heat is transferred by the process of convection from the gas flame at the bottom of the oven to the potatoes at the top of the oven.

Model Answer

The air particles are heated by the gas flame and gain energy. This causes the particles to move faster and spread out. Because the particles are spread out the hot air becomes less dense and rises.



Real world Examples

A lava lamp - The light bulb heats the wax causing it heat up and rise. The wax cools at the top and sinks again.

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Conduction and Convection

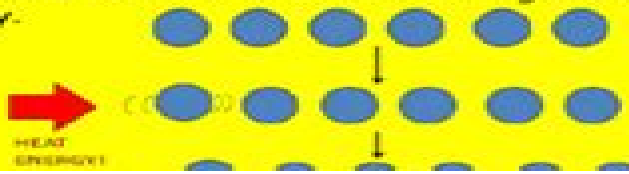
Conduction

Definition

Conduction is how heat energy is transferred through solids when they are heated. Heat energy can also be passed from one solid to another by conduction.

Conductors are materials which transfer energy more easily. Insulators are materials which don't transfer heat as well for example, Glass and plastics.

As the conductor is heated the atoms gain more energy and vibrate more. This causes them to collide with other atoms transferring the heat energy.



Model Question

Explain in terms of particles how heat is transferred through the base of the pan?



Model Answer

Atoms in the base of the saucepan gain thermal energy from the hob. This causes the atoms in the saucepan to vibrate. Due to the vibration the atoms will collide with other atoms and pass on their thermal energy.

Why are metals good conductors?

In a metal lattice electrons from the outer shell are freed from their atoms. This causes those atoms to become positively charged ions. As the metal is heated, these ions vibrate more, passing on their energy to the electrons, which then move through the metal, carrying the heat energy.

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Marcel A. Müller



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1. Understanding the eBook Pixl Science P1
 - The Rise of Digital Reading Pixl Science P1
 - Advantages of eBooks Over Traditional Books
2. Identifying Pixl Science P1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pixl Science P1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pixl Science P1
 - Personalized Recommendations
 - Pixl Science P1 User Reviews and Ratings
 - Pixl Science P1 and Bestseller Lists
5. Accessing Pixl Science P1 Free and Paid eBooks

- Pixl Science P1 Public Domain eBooks
- Pixl Science P1 eBook Subscription Services
- Pixl Science P1 Budget-Friendly Options
- 6. Navigating Pixl Science P1 eBook Formats
 - ePub, PDF, MOBI, and More
 - Pixl Science P1 Compatibility with Devices
 - Pixl Science P1 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pixl Science P1
 - Highlighting and Note-Taking Pixl Science P1
 - Interactive Elements Pixl Science P1
- 8. Staying Engaged with Pixl Science P1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pixl Science P1
- 9. Balancing eBooks and Physical Books Pixl Science P1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pixl Science P1
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pixl Science P1
 - Setting Reading Goals Pixl Science P1
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pixl Science P1
 - Fact-Checking eBook Content of Pixl Science P1
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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