

LAB: CALCULATING A LIMITING REACTANT

Name _____

Lab Determination of a Limiting Reactant

Objective: Observe chemical reaction between magnesium metal and hydrochloric acid. Calculate the limiting and excess reactant.

Materials: 2 M HCl test tube

Procedure:

1. Record the mass of magnesium: _____ g
2. Place the test tube in a rack. Add the magnesium strip.
3. Add 4 mL of 2 M HCl into the test tube.
4. Record your observations. How do you know a chemical reaction is taking place? Be sure to feel the side of the test tube.
5. Pour the contents into the waste beaker provided by your teacher. Rinse the rack upside down. You never want to pour unreacted solid.
6. Write the balanced equation for the reaction that you observed and balance the equation.



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M Mosston



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