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Titration

Answer Key

Vocabulary: acid, analyte, base, dissociate, equivalence point, indicator, litmus paper, molarity, neutralize, pH, strong acid, strong base, titrant, titration, titration curve, weak acid, weak base

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)
[Note: The purpose of these questions is to activate prior knowledge and get students thinking. Students are not expected to know the answers to the Prior Knowledge Questions.]

There are several definitions of acids and bases. According to the Bronsted-Lowry definition, an **acid** is a substance that is capable of donating a proton to another substance. A **base** is a substance that accepts protons. When an acid and a base are combined, the acid is **neutralized** as the base accepts the protons produced by the acid.

One way to determine if a solution is acidic or basic is to use **litmus paper**, as shown above. There are two types of litmus papers: red and blue.

How does litmus paper indicate an acid? *Both strips turn red.*

How does litmus paper indicate a neutral substance? *The red strip stays red and the blue strip stays blue.*

How does litmus paper indicate a base? *Both strips turn blue.*

Gizmo Warm-up
Litmus is an example of an **indicator**, a substance that changes color depending on its **pH** (pH is a measure of the concentration of protons, or H^+ ions). In the *Titration* Gizmo, you will use indicators to show how acids are neutralized by bases, and vice versa.

To begin, check that **1.00 M NaOH** is selected for the **Burette**, **Mystery HBr** is selected for the **Flask**, and **Bromthymol blue** is selected for the **Indicator**.

1. Look at the flask. What is the color of the bromthymol blue indicator? *Yellow*

2. What does this tell you about the pH of the solution in the flask? *pH is below 6.0*

   Solutions with a pH below 7.0 are acidic, while those with a pH above 7.0 are basic.

3. Move the slider on the burette to the top to add about 25 mL of NaOH to the flask. What happens, and what does this tell you about the pH of the flask?

   The solution turns blue, indicating the pH is now above 7.6.

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Julia Schneider

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Table of Contents Quick Titration Lab Answers

1. Understanding the eBook Quick Titration Lab Answers
    - The Rise of Digital Reading Quick Titration Lab Answers
    - Advantages of eBooks Over Traditional Books
2. Identifying Quick Titration Lab Answers
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Quick Titration Lab Answers
    - User-Friendly Interface
4. Exploring eBook Recommendations from Quick Titration Lab Answers
    - Personalized Recommendations
    - Quick Titration Lab Answers User Reviews and Ratings
    - Quick Titration Lab Answers and Bestseller Lists

5. Accessing Quick Titration Lab Answers Free and Paid eBooks
    - Quick Titration Lab Answers Public Domain eBooks
    - Quick Titration Lab Answers eBook Subscription Services
    - Quick Titration Lab Answers Budget-Friendly Options
6. Navigating Quick Titration Lab Answers eBook Formats
    - ePub, PDF, MOBI, and More
    - Quick Titration Lab Answers Compatibility with Devices
    - Quick Titration Lab Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Quick Titration Lab Answers
    - Highlighting and Note-Taking Quick Titration Lab Answers
    - Interactive Elements Quick Titration Lab Answers
8. Staying Engaged with Quick Titration Lab Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Quick Titration Lab Answers
9. Balancing eBooks and Physical Books Quick Titration Lab Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Quick Titration Lab Answers
10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
11. Cultivating a Reading Routine Quick Titration Lab Answers
    - Setting Reading Goals Quick Titration Lab Answers
    - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Quick Titration Lab Answers
    - Fact-Checking eBook Content of Quick Titration Lab Answers
    - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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