Significant Figure and Scientific Notation Worksheet

1. Convert each of the following into scientific notation.

a) 3427

ь) 0.00456

c) 123, 453

d) 172

e) 0.000984

0.502

g) 3100.0×10^{2}

b) 0.0114×104

i) 107.2

j) 0.0000455

k) 2205.2

1) 30.0×10^{-2}

m) 0.982×10^{-3}

n) 0.0473

0) 650,502

p) 3.03×10^{-1}

 $q)20.4 \times 10^{5}$

r) 1000×10⁻³

2. Determine the number of significant digits in each of the following:

a) 3427

b) 0.00456

c) 123,453

d) 172

c) 0.000984

0.502

g) 3100.0×10²

b) 0.0114×104

i) 107.2

j) 0.0000455

k) 2205.2

1) 30.0×10^{-2}

m) 0.982×10^{-3}

n) 0.0473

o) 650,502

p) 3.03×10^{-1}

q) 20.4×10^5

r) 1000×10⁻³

3. Convert each into decimal form.

a) 1.56×10^4

b) 0.56×10^{-2}

c) 3.69×10^{-2}

d) 736.9×10⁵

e) 0.00259×10⁵

f) 13.69×10^{-2}

g) 6.9×10^4

Scientific Figures Mastering Chemistry Answers

David Kirk

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